

**High-precision ID-TIMS Cassiterite U-Pb systematics using a low-contamination hydrothermal decomposition: implications for LA-ICP-MS and ore deposit geochronology**

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## 10 1. Abstract

Cassiterite ( $\text{SnO}_2$ ) is the most common ore phase of Sn. Typically containing 1-100  $\mu\text{g/g}$  U and relatively low concentrations of common Pb, ~~cassiterite the mineral~~ has been increasingly targeted for U-Pb geochronology, principally via micro-beam methods, to understand the timing and durations of granite related magmatic-hydrothermal systems throughout geological time. However, due to the extreme resistance of cassiterite to most forms of acid digestion, ~~to date,~~ there has been no published method permitting the complete, closed system decomposition of cassiterite under conditions where the basic necessities of measurement by isotope dilution can be met, leading to a paucity of reference and validation materials. To address this a new low blank ( $<1$  pg Pb) method for the complete acid decomposition of cassiterite utilising HBr in the presence of a mixed U-Pb tracer, U and Pb purification, and TIMS analyses has been developed. Decomposition rates have been experimentally evaluated under a range of conditions. A careful balance of time and temperature is required due to competing effects (e.g. HBr oxidation) yet decomposition of 500  $\mu\text{m}$  diameter fragments of cassiterite is readily achievable over periods comparable to zircon decomposition. Its acid resistant nature can be turned into an advantage, by leaching common Pb-bearing phases (e.g. sulfides, silicates) without disturbing the U-Pb systematics of the cassiterite lattice. The archetypal Sn-W greisen deposit of Cligga Head, SW England, is used to define accuracy relative to CA-ID-TIMS zircon U-Pb ages and demonstrate the potential of this new method, for resolving high resolution timescales ( $<0.1$  %) of magmatic-hydrothermal systems. However, ~~analyses data~~ also indicate that the isotopic composition of initial common Pb varies significantly, both between crystals and within a single crystal. This is attributed to significant fluid-rock interactions and the highly F-rich acidic nature of the hydrothermal system. At micro-beam precision levels, this issue is largely unresolvable and can result in significant inaccuracy in interpreted ages. ~~However,~~ this method can, for the first time, be used to properly characterise suitable reference materials for micro-beam cassiterite U-Pb analyses, thus improving the accuracy of the U-Pb cassiterite chronometer as a whole.

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## 2. Introduction

Cassiterite ( $\text{SnO}_2$ ) is the primary tin ore, precipitated from magmatic-hydrothermal fluids exsolved in association with reduced granitic systems and pegmatite bodies. The mineral systems that contain cassiterite are globally distributed and they span the majority of geological time from the Archean to the Neogene (Kesler and Wilkinson, 2015). Cassiterite commonly contains 1 to 100  $\mu\text{g/g}$  U and has the potential to contain relatively low levels of common Pb (e.g. Li et al., 2016; Moscati and Neymark, 2019; Neymark et al., 2018). The potential of cassiterite as a U-Pb geochronometer to understand the timing of magmatic-hydrothermal Sn deposits, and their many associated elements critical for modern technology e.g. Li, W, Nb, Ta, has long been recognised since the first reported Isotope Dilution Thermal Ionisation Mass Spectrometry (ID-TIMS) U-Pb analyses using (Gulson and Jones, 1992). ~~Yet~~ ~~Research~~ into cassiterite U-Pb geochronology paused for some time due to the questionable accuracy of the first published results (McNaughton et al., 1993) and the difficulty in the hydrothermal decomposition of cassiterite required for isotope dilution methods due to the extremely acid resistant nature of cassiterite- (Gulson and Jones, 1992; Neymark et al., 2018).

With the advent of micro-beam techniques, cassiterite U-Pb geochronology has now become widely accessible. Techniques such as LA-ICP-MS and Ion microprobe (Carr et al., 2017) offer a rapid solution to analysing the U-Pb systematics of cassiterite without the direct need for dissolving a sample.

With analytical precision of  $\sim 1-2\%$   $2\sigma$ , and a current total uncertainty limit of  $\sim 2\%$  based upon long-term reproducibility and inter-laboratory comparison of zircon U-Pb analyses (Horstwood et al., 2016), the LA-ICP-MS U-Pb techniques offer the capability to provide chronology that may address issues such as the timing of ore deposition at broad, regional scales. Issues specific to cassiterite LA-ICP-MS U-Pb analyses are outlined below. Higher temporal resolution questions, such as those that begin to address processes on magmatic timescales require a refinement of these uncertainties. For micro-beam techniques this may result from the refined standardisation of the methodology and identification of systematic uncertainties identified over long term analysis using well characterised reference materials (RMs)(Horstwood et al., 2016).

Accuracy of U-Pb micro-beam geochronology requires that effects such as matrix interaction with analyte ion beams and “down-hole” inter-element fractionation can be corrected for. This is commonly  
65 negated by using RMs with an equivalent chemistry and ablation (or sputtering) characteristics (i.e. a matrix matched RM). Samples of unknown isotopic composition are measured relative to well characterised RMs that are relatively homogenous at the scale of the spatial resolution and at the precision of the analyses and typically characterised by ID-TIMS U-Pb analyses.→

Thus far, a fundamental issue for cassiterite U-Pb geochronology arises from the challenges in  
70 achieving total decomposition of cassiterite, in a single closed system step, that is required for the effective quantitative U/Pb determination of the material by isotope dilution methods. Previous ID-TIMS U-Pb studies (Gulson and Jones, 1992; Rizvanova et al., 2017; Yuan et al., 2008) utilising concentrated HCl required stepwise decomposition and acid recharges, that still only achieved partial decompositions in many instances. These studies also required the post-decomposition addition of  
75 isotopic U-Pb tracers. This approach will likely preserve the Pb isotope systematics of the sample, for cassiterite this will represent a mixture of common Pb (Pbc) components and radiogenic Pb (Pb\*). However, it introduces a high potential to fractionate the U-Pb inter-element ratios, either due to the incongruous dissolution of U and Pb, and different elemental behaviours before equilibrium with the tracer is achieved. These issues will in turn affect the integrity of U-Pb derived dates and therefore the  
80 interpretation of Pb loss, or Pbc-Pb\* mixing trajectories that require assumptions about concordance of the U-Pb chronometers. As noted by Neymark et al. (2018) these effects lead to reverse discordance (U loss or Pb gain) between the resulting cassiterite analyses. Whilst this effect can be easily detected in older cassiterite data where trajectories approach being perpendicular to Concordia, within younger samples, fractionation will merely drive samples near parallel to the line of Concordia, making it  
85 difficult to distinguish from real age variations.

Many previous micro-beam cassiterite U-Pb studies have used some combination of zircon RM or glass RM, with a “matrix matched” cassiterite RM that has had isotopic ratios characterised by ID-TIMS following a stepwise decomposition (e.g. Yuan et al., 2011, 2008; Zhang et al., 2017; Li et al., 2016;

Liu et al., 2007). The reference values have the potential to have been effected by the issues described above, with inaccuracies propagating into the resulting in-situ ages.

~~Neymark et al. (2018) developed an approach that utilised the highly repeatable ablation characteristics~~

~~of cassiterites to overcome the matrix matching issues in micro-beam analysis that stem from the absence of well characterised cassiterite material that can be used as a RM.~~

~~Neymark et al. (2018) developed an approach that utilised the highly repeatable ablation characteristics of cassiterites, to~~

~~normalise the matrix biases relative to a NIST glass primary RM were normalised the~~ by deriving

a “fractionation factor” for each analytical session. This was achieved by measuring the offset between

the lower intercept age on a Terra Wasserburg (T-W) plot, and an assumed “true” age defined by the

Pb-Pb isochron that was shown to be in good agreement with independent chronological constraints for

the Proterozoic “SPG” cassiterite. The application of this approach to a wide range of materials across

geological time, and the general agreement with independent geological and temporal constraints of

their associated systems (Moscati and Neymark, 2019; Neymark et al., 2018) indicates that this

approach was both versatile, and can be reliably applied to constrain the general timing of ore deposits.

However, without characterising ~~the U-Pb systematics of cassiterite materials by precise methods, and~~

~~understanding, the U-Pb systematics of cassiterite materials by precise methods, the potential level of~~

~~accuracy in resulting geological interpretations of the methods, and the potential interpretive resolution~~

~~for geological problems~~ are difficult to assess.

Regardless of the micro-beam approach adopted, it is crucial that the session and long term accuracy

and dispersion of any U-Pb analytical approach can also be validated by the analysis of independently

well characterised cassiterites (i.e. secondary and tertiary RMs) (e.g. Horstwood et al. 2016).

The motivation for developing a method for cassiterite ID-TIMS U-Pb geochronology that involves the

total decomposition of cassiterite in the presence of, and achieving equilibrium with a U-Pb isotope

tracer, is to provide a means to characterise cassiterite materials to assess accuracy, and a means to

examine magmatic-hydrothermal timescales within cassiterite bearing deposits beyond the consensus

ca. 2 % absolute uncertainty that is permitted by micro-beam U-Pb methods (e.g., Horstwood 2016).

115 Desirable characteristics of a method are also: 1) contribution of low amounts of Pb to the sample during decomposition and chemical separation using reagents that can be easily handled and distilled; 2) ~~that~~ the method can achieve decomposition on timescales that are easily operable for the laboratory environment e.g. zircon decomposition is typically achieved within ~72 hours; 3) ~~That~~ the method utilises geologically meaningful amounts of material, capable of resolving spatial variations in isotopic ratios.

This study demonstrates the potential of cassiterite decomposition with concentrated HBr to fulfil the necessary criteria for routine ID-TIMS cassiterite U-Pb geochronology. We provide preliminary data for two cassiterite materials that have the potential for use as RMs, and we evaluate the accuracy of cassiterite U-Pb geochronology using the case study of the classic W-Sn magmatic hydrothermal system of Cligga Head, SW England, which illustrates the complexities and biases that may arise for cassiterite U-Pb geochronology.

### 3. Previous methods for the decomposition of Cassiterite

Cassiterite is notoriously resistant to decomposition in acids (Mathur et al., 2017; Yamazaki et al., 2013). Although analysis of Sn isotopes within cassiterite is well established, decomposition has typically been conducted by alkali fusion (Hall, 1980; Sear, 1997), reduction with graphite (Clayton et al., 2002; Hall, 1980) or potassium cyanide reduction (Haustein et al., 2010; Mathur et al., 2017). These methods require high temperatures (800-1200°C) and open systems, making them inappropriate for ID U-Pb geochronology.

Hydroiodic acid has been demonstrated as an effective means of cassiterite decomposition at relatively low temperatures 100°C (Caley, 1932). More recent work was able to dissolve 1 mg of cassiterite in high pressure decomposition vessels at 100°C over four sequences of 24 hours, exchanging the acid at each step (Yamazaki et al., 2013; Mathur et al., 2017). Despite the potential of HI as a means to decompose cassiterite, there are several drawbacks as HI readily reacts with oxygen within air requiring refrigerated storage, ~~and at the time of writing was~~ is less readily available with certification to <1 pg/ml Pb, and ~~was~~ is significantly more expensive when compared with HBr. ~~Likewise~~ Similarly, the

decomposition procedure of Yamazki et al. (2013) yielded a 1 ng Sn blank, and whilst it is conjecture to predict the associated Pbc blank, if on this order of magnitude it would rule out HI as a low contamination method for high-precision geochronology of cassiterite. ~~Furthermore, u~~ Upon drying of the dissolved samples Yamazki et al. (2013) noted the formation of precipitates which were insoluble with a range of acids (HCl, HF, HClO<sub>3</sub>, HNO<sub>3</sub>), it is possible that these could also sequester Pb and U, significantly decreasing recoveries.

Previous efforts to analyse cassiterite U-Pb through ID methods for geochronology have utilised a concentrated HCl based decomposition within high-pressure acid digestion vessels (Gulson and Jones 1992; Liu 2007; Yuan et al., 2008; Yuan et al., 2011; Rizvanova et al., 2017). The previously reported contributions of Pbc from the method have been ~~have been~~ on the order of 0.2-0.3 ng (Gulson and Jones, 1992) to 0.8 ng (Yuan et al., 2011, 2008), many orders of magnitude greater than what is typically achieved by modern zircon CA-ID-TIMS geochronology methods (<0.5 pg). The issues of incomplete decomposition in previous studies was highlighted by Neymark et al. (2018). Although not explicit about the number of fractions that did not dissolve, the methods of Yuan et al. (2008) indicated that at least some of their cassiterite was not decomposed after 72 h in 12 M HCl at 205°C in high-pressure digestion vessels, and required multiple acid exchange steps, and also further crushing with a pestle and mortar in-between decomposition steps. Gulson and Jones (1992) also utilised a 48-72 h, ~200°C, 12 M HCl decomposition stage within a high pressure digestion vessel and noted that in some instances for ~200 Ma cassiterite, <5 % of cassiterite was dissolved after numerous steps.

The potential for inaccuracy in cassiterite U-Pb geochronology was illustrated by ~~the study of~~ Gulson and Jones (1992), who presented a very precise age ( $\pm \sim 0.1\% 2\sigma$ ) from a discordia regression line with an upper intercept age of 2098.6 Ma  $\pm$  3.1 Ma for cassiterite associated with granitic rocks within the Bushveld Complex, South Africa. This cassiterite age was quickly challenged at the time on account of much younger zircon ages of the associated granites (McNaughton et al., 1993). Extensive high-precision CA-ID-TIMS zircon U-Pb geochronology of the relatively older Rustenberg layered igneous suite of the Bushveld Complex (Mungall et al., 2016; Scoates and Friedman, 2008; ~~Scoates and Wall.~~

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2015; Zeh et al., 2015) has now confidently demonstrated that the cassiterite age is at least > 40 Myrs (>2 %) too old.

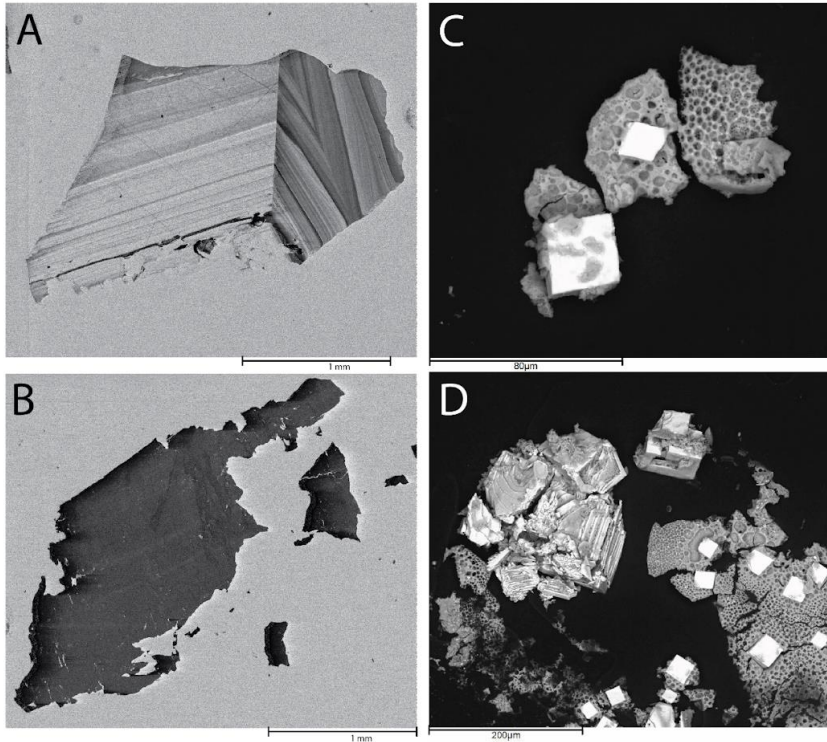
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170 The potential to decompose tin oxides at pressure using 9 M (48 %) HBr in high pressure digestion vessels has been identified for some time now (Doležal et al., 1969), yet its application has ~~yet~~ been somewhat overlooked for U-Pb geochronology despite its strong acidic, reducing nature that is comparable to HI.

#### 175 **4. Controls on the timescales of cassiterite hydrothermal decomposition with 9M HBr**

The ability of 9 M HBr to decompose Cassiterite over a range of temperatures and grain sizes was investigated in a series of step-leaching experiments. Timescales were assessed for two different cassiterite crystal samples (Fig. 1): 1) “Casnig” with an age of ~150 Ma and a heavily quenched Cathodoluminescence (CL) response, indicating a high concentration of Fe (Farmer et al., 1991); 2) A  
180 single ~1 cm diameter cassiterite from the Permian Cligga Head, SW England which demonstrates strong zonation under CL. Around 0.015 g to 0.03 g of cassiterite fragments, both powdered in a pestle and mortar to <10 µm to 50 µm, and as coarse 500 µm length cubes, were decomposed in a 50x molar excess of 9 M HBr in Savillex high pressure digestion vessels (Parr Bombs) at a range of temperatures 180°C-230°C. Decomposition steps lasted for 12 h, or 96 h, and the mass decrease in the residual  
185 cassiterite was documented (see supplementary material for full data), following extraction and extensive rinsing of the leachate. The acid was recharged and the experiment repeated. Complexity in the weighing arose from Br<sub>2</sub> released during the decomposition steps, diffusing into the PFA vessels thus increasing the mass. The mass would then decrease as Br<sub>2</sub> diffused out of the PFA vessel during hot plate dry down steps (120°C) before weighing. Experimental runs of PFA vessels without cassiterite  
190 present indicated the potential mass variation at each weighing stage to be ±0.00154 g (2s). Scanning Electron Microscope imagery and Energy Dispersive Spectroscopy of the precipitated leachate after a decomposition step identified crystalline tin bromide salts, and spongy textured Ti, Nb and Fe bromides which reflect minor elements contained within the cassiterite (Fig. 1). No tin oxide was identified supporting that the mass reduction at each stage was due to decomposition of cassiterite in HBr.





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**Figure 1.** A) Representative CL image of cassiterite from Cligga Head used for decomposition experiments and geochronology; B) Representative CL image of the Casnig cassiterite used for decomposition experiments; C and D) SEM images of the precipitate formed after drying down the HBr leachate after cassiterite decomposition. Light coloured crystals are tin bromide crystals, the “spongy” texture contains bromides of Fe, Nb and Ti.

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Decomposition experiments run at 180°C and 200°C did not exhibit a significant reduction in mass beyond the mass variation in the beaker due to Br<sub>2</sub> diffusion. No clear temperature control was exerted between 210°C and 230°C and most experiments demonstrated a relatively large variation in the amount dissolved over 12 h from step to step. Likewise, there was no consistent variation in the amount decomposed over 12 h between the two different starting materials of cassiterite at a given temperature.

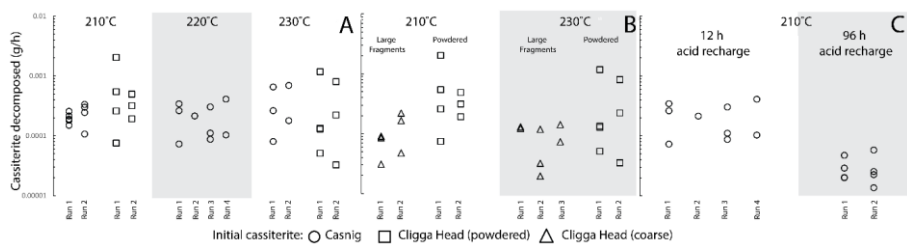
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In general more cassiterite decomposed when powdered in a pestle and mortar than when placed in as larger fragment, although significant variation in the rate of a given step was still present.

210 Notably there was little variation between the total mass of cassiterite decomposed in either a 96 h step or a 12 h step, indicating the majority of cassiterite decomposition occurs within the first 12 h or less. We suggest this apparent decrease in rate resulted from a rapid reduction in HBr molarity when HBr was exposed to high temperatures. The strong brown discolouration and mass increase of the decomposition vessel that was not unique to cassiterite bearing experiments, implied that Br<sub>2</sub> was  
215 generated likely through a reaction with atmospheric oxygen. It is well documented that aqueous solutions of HI readily react with atmospheric oxygen at room temperatures to produce I<sub>2</sub>, which further react with HI to produce HI<sub>3</sub>, and a similar reaction between O<sub>2</sub> and HCl occurs, just at a much slower rate (and requiring higher temperatures). It follows that this reaction occurs with HBr, although it appears to not be previously well documented.

220 The decomposition experiments act as an informative guide to cassiterite decomposition, but given the scales of the experiment design required to overcome weighing issues, it is unlikely they comprehensively reflect the rates of cassiterite decomposition in 9M HBr. Whether a cassiterite sample completely decomposes within a single step is likely to be fundamentally controlled by the surface area to volume (or mass) ratio of the cassiterite that is exposed to the acid during the earliest stages of  
225 decomposition, before the HBr is lost and becomes ineffective for cassiterite decomposition. Use of a significant excess of concentrated HBr is proposed, in the later experiments for U-Pb geochronology a molar excess of ~190 times was used. To ensure complete single step decomposition can be achieved, we advise carrying out a trial decomposition experiment to ensure a given mass and fragment size of a specific cassiterite sample can be decomposed within the parameters of the experiment chosen.

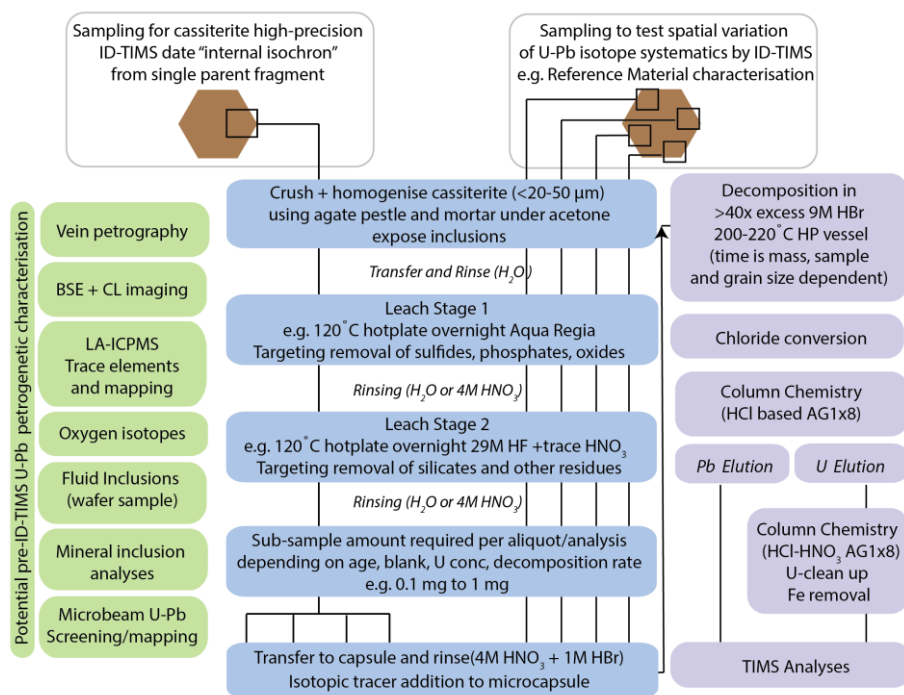
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**Figure 2.** Plots of integrated decomposition rates for different initial start cassiterite materials. Variations in mass below those induced by  $\text{Br}_2$  diffusion into the beaker are not displayed A) Shows variation of rate over a 12 h decomposition step at different temperatures within high pressure vessels for powdered cassiterite. B) Shows differences in rates between coarse fragments approximating cubes 500  $\mu\text{m}$  in length, and cassiterite powdered in a pestle and mortar at 210°C and 230°C. C) Illustrates the difference in rate averaged over 12 h to 96 h decomposition steps at 210°C. See text for discussion.

## 240 5. Method for Cassiterite ID-TIMS U-Pb geochronology

An overview of the methods employed and described within this section is provided in Figure 3, in addition to petrogenetic characterisation techniques that can be employed within the workflow prior to ID-TIMS U-Pb analyses, although these have not been fully utilised within this study. The specifics of sampling style, sizes and experimental conditions are offered as guidelines. Their selection will be dependent on user requirements and the nature of the sample material.



**Figure 3.** A generalised workflow for cassiterite ID-TIMS U-Pb geochronology. ID-TIMS analyses can be employed after the listed petrogenetic characterisation techniques on sample material. Schematic representation of sample aliquots and sub-aliquots illustrates the different approaches utilised within the study based on application (characterising heterogeneity in “reference materials” or internal isochron approach to avoid issues with spatial heterogeneity in Pb composition) see Fig.7 for further details on “internal isochron” formed from sub-aliquots. Note many experimental variables will be sample material and user requirement dependent. For discussion of decomposition times see Section 4.

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### 255 5.1 Sample preparation

Fragments of cassiterite grains were crushed under acetone in a pre-cleaned agate pestle and mortar, and transferred to a PFA beaker. The acetone was then evaporated and the powdered material was rinsed first in H<sub>2</sub>O, then with 4 M HNO<sub>3</sub> and before being refluxed in aqua regia at 120°C overnight

with the aim of dissolving sulfide and Fe-oxides inclusions that were exposed ~~to the surface of the~~ within  
260 the powdered cassiterite grains (e.g. Gulson and Jones, 1992). ~~The samples~~ was/were rinsed in 4 M  
HNO<sub>3</sub> and H<sub>2</sub>O, and i. In the case of samples that underwent an additional step of HF leaching samples  
were, refluxed at 120°C overnight in ~300-400 µl 29 M HF and trace HNO<sub>3</sub> on a hot plate. The leached  
cassiterite was then rinsed in 4 M HNO<sub>3</sub> and H<sub>2</sub>O. ~~Sub-~~ aliquots of leached cassiterite material that  
265 visually ~~approximating~~ approximated the material contained in a ~500 µm sided cubes ~~of the leached~~  
~~powdered~~ of cassiterite were then transferred into 500 µl PFA microcapsules (Parrish, 1987) that had  
undergone several steps of overnight cleaning in 9 M HBr within the Parr vessels, in addition to normal  
HF and HCl cleaning routines. Samples were briefly (~30 minutes) refluxed (hotplate, 120°C) and  
rinsed in 1 M HBr, leaving a minimal amount of acid in the capsule, whilst at the same time ensuring  
no sample material was removed before being spiked with ~0.005 g of EARTHTIME 535 tracer  
270 (Condon et al., 2015). Approximately 450 µl of 9 M HBr (Romil Ultra Purity Acid, UPA analysed to  
have <0.04 pg/ml Pbc) was added before placing in the oven for 5 nights at 210°C within a high pressure  
digestion vessel. Following decomposition, the solution was dried on a hot plate at 120°C and converted  
to a chloride by the addition of 400 µl of 3 M HCl and placing in the oven for a further 12 hours at  
180°C using a high pressure vessel.

275 One noteworthy observation made over the course of cleaning the PFA capsules in 9M HBr overnight  
at 230°C was that the apparent Pbc was found to have doubled relative to the previous cleaning stage  
at 210°C, and the equivalent amount of in-house <sup>208</sup>Pb tracer produced very poor beam intensities. This  
was interpreted as a result of organic compounds released by the reaction of the HBr with the PFA  
capsule. Despite potential benefits from more rapid decomposition at 230°C, lower temperatures are  
280 recommended.

## 5.2 Cassiterite U-Pb anion exchange chromatography

A two-stage column chemistry procedure was determined f following investigations to isolating-isolate  
Pb and U from matrix elements using 0.05 ml of pre-cleaned AG 1-x8 200-400 mesh resin (BIO-RAD,  
285 CA, USA) and a homogeneous solution of dissolved cassiterite equating to a ~500 µm length cube grain

(or 0.875 mg at an assumed density of  $\sim 7 \text{ g/cm}^3$ ) ~~and 0.05 ml of pre-cleaned AG 1-x8 200-400 mesh resin (BIO-RAD, CA, USA), a two stage column chemistry procedure was determined.~~ In the first stage a modification of the HCl based anion exchange chromatography typically used for zircons (Krogh, 1973) was used. The sample was loaded onto the columns in 3 M HCl, before further 3M HCl being added in several stages of 30  $\mu\text{l}$  and then several stages of 130  $\mu\text{l}$  3 M HCl. These washes contained the majority of eluted Nb, Ti and Th. The Pb was eluted in three stages of 150  $\mu\text{l}$  6M HCl, collected, and dried down with  $\sim 10 \mu\text{l}$  0.03 M  $\text{H}_3\text{PO}_4$ . The U was eluted using three steps of 150  $\mu\text{l}$  of  $\text{H}_2\text{O}$ , followed by one step of 150  $\mu\text{l}$  of 1M HCl. As the U fraction was found to contain almost all eluted Fe and Sn, a further U clean up stage was required. The U-bearing solution was dried and re-dissolved in 300  $\mu\text{l}$  8 M  $\text{HNO}_3$  before being loaded on to columns, again containing 0.05 ml of pre-cleaned AG 1-x8 resin. Columns were washed in three stages of 350  $\mu\text{l}$  8 M  $\text{HNO}_3$  followed by two steps of 350  $\mu\text{l}$  8 M HCl. The U was eluted using 2 steps of 200  $\mu\text{l}$  0.2 M HCl, collected, and dried down with  $\sim 10 \mu\text{l}$  0.03 M  $\text{H}_3\text{PO}_4$ .

~~For steps after crushing in a pestle and mortar (The Pbc procedural blank, for steps after crushing in a pestle and mortar,~~ was found to be on the order of  $\sim 0.5 \text{ pg}$ . The procedural blank will be subject to laboratory variations, and the efficacy with which any material abraded during crushing within a pestle and mortar can be removed prior to decomposition within HBr. The effects of HF leaching prior to decomposition are evaluated below, the lowest total sample and laboratory Pbc of  $\sim 1.5 \text{ pg}$  including any pestle and mortar contributions indicates that crushing is a minor contributor post-HF leaching. Crucially the component of Pbc contributed by the method is one to four orders of magnitude less than the amount of  $\text{Pb}^*$  of the materials examined within this study, and the initial Pbc within the cassiterite is likely one or two orders of magnitude greater than that from the laboratory.

### 5.3 Mass spectrometry and data reduction

The Pb and U of a given sample were independently loaded on a zone-refined Re filament in 1.5  $\mu\text{l}$  of silica gel matrix (Gerstenberger and Haase, 1997). Isotope ratio measurements were made using a Thermo Triton TIMS at the British Geological Survey following typical methods described by Tapster

et al. (2016). Raw U and Pb data were filtered using the Tripoli software program (Bowring et al., 2011). Data reduction and uncertainty propagation used a modified excel spreadsheet (Schmitz and Schoene, 2007) with the U decay constants of Jaffey et al., (1971), the  $^{238}\text{U}/^{235}\text{U}$  ratio of Hiess et al. (2012) and the decay constants of Cheng et al., (2000) for  $^{230}\text{Th}$ . Concordia diagrams and regressions were constructed using the Excel add-in 'ISOPLOT 4.15' (Ludwig, 2008) and initial disequilibrium corrections utilised the IsoplotR package (Vermeesch, 2018)

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## 6. Cassiterite ID-TIMS U-Pb results

### 6.1 SPG-IV ad hoc RM Cassiterite

The SPG-IV cassiterite is taken from the Pitkäranta ore district, Russian Karelia, and was utilised by Neymark et al. (2018) as a cassiterite reference material for deriving the "fractionation factor" of analytical sessions. The age of the cassiterite based on geological constraints and zircon dating of associated magmatism was inferred between 1546.7 to 1537.9 Ma (Amelin et al., 1997). Neymark et al. (2018) presented a LA-ICP-MS  $^{207}\text{Pb}^*/^{206}\text{Pb}^*$  weighted mean date of  $1542.7 \pm 1.5$  Ma and a date of  $1539.8 \pm 3.5$  Ma for three partially dissolved, reversely discordant ID-TIMS analyses (N. Rizvanova, written communication, 2017 in Neymark et al. 2018).

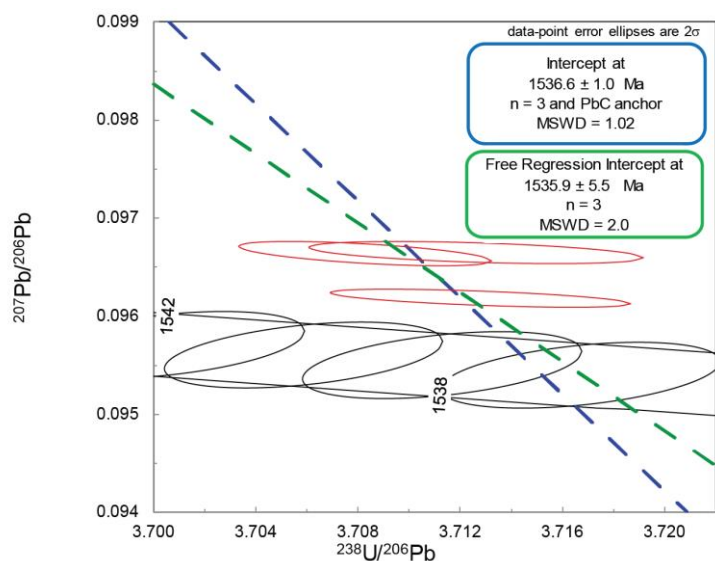
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The free regression of three tightly clustered ID-TIMS analyses (Fig. 34) of spatially independent aliquots of the SPG-IV cassiterite yields a lower intercept of  $1535.9 \pm 5.5$  Ma (MSWD = 2.0). As utilised by Neymark et al. (2018) for dates that cluster close to Concordia, anchoring the common Pb component of the T-W isochron to the mean  $^{207}\text{Pb}/^{206}\text{Pb}$  value of 11 galena analyses from the ore system ( $1.0104 \pm 0.0062$ ) (Larin et al., 1990) generates a lower intercept date of  $1536.6 \pm 1.0$  (MSWD = 1.02) (Fig. 4). The total Pbc amounts for these analyses ranges from ~5 pg to 15 pg and  $^{206}\text{Pb}/^{204}\text{Pb}$  range from ~12000 to ~23000.

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Comparison of the ID-TIMS T-W lower intercept dates with the LA-ICP-MS Pb-Pb isochron dates used to derive the fractionation factor (Neymark et al., 2018) indicates an absolute offset ~0.4 % older for the LA-ICP-MS Pb-Pb isochron data. The equivalent ID-TIMS  $^{204}\text{Pb}/^{206}\text{Pb}$ - $^{207}\text{Pb}/^{206}\text{Pb}$  isochron date for

340 this study yields  $1540.9 \pm 3.6$  Ma that is broadly in concert with that derived by the LA-ICP-MS data (Neymark et al., 2018).



345 **Figure 34.** T-W plot of SPG-IV cassiterite showing lower intercept dates for freely regressed data (green line), and regression anchored to the Pbc  $^{207}\text{Pb}/^{206}\text{Pb}$  of Galena analyses from the ore system (blue line).

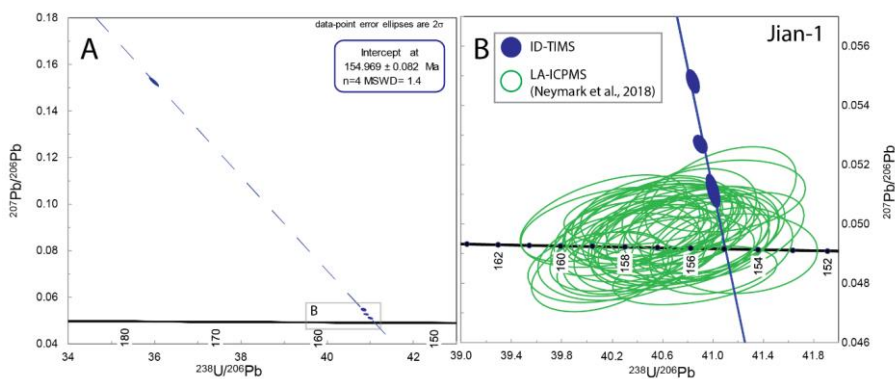
## 350 6.2 Jian-1 Cassiterite

The Jian-1 cassiterite is derived from the Jiangxi W-Sn district, South China (Neymark et al., 2018). Previously a LA-ICP-MS cassiterite weighted mean  $^{206}\text{Pb}/^{238}\text{U}$  date of  $159.5 \pm 1.5$  Ma ( $n = 31$ ; MSWD = 0.4) (Zhang et al., 2017) was reported for the deposit. Superseded by a weighted mean  $^{206}\text{Pb}/^{238}\text{U}$  date of  $156.55 \pm 0.36$  Ma ( $n = 40$ ; MSWD=1.4) (Neymark et al. 2018). In both studies data points  
355 overlapping Concordia within uncertainty and are interpreted as concordant, or free of Pbc.



Four individual aliquots of the Jian-1 Cassiterite were analysed by ID-TIMS. All data are discordant on the T-W plot (Fig. 45), and are interpreted as containing a component of initial common Pb. The total Pbc amounts for these analyses ranges from ~1.5 pg to 14 pg and their  $^{206}\text{Pb}/^{204}\text{Pb}$  range from ~140 to ~5300. The three data points with greatest  $^{206}\text{Pb}/^{204}\text{Pb}$  form a lower intercept of  $155.05 \pm 0.20$  Ma (MSWD = 2.1), in agreement with the lower intercept formed by all four data points of  $154.969 \pm 0.082$  Ma (MSWD = 1.4).

Comparison with LA-ICP-MS data (Fig. 45) indicate the coarser-scale sampling of ID-TIMS analyses identified domains with lower Pb\*/Pbc relative to the previously published LA-ICP-MS data (Neymark et al., 2018) and therefore provide additional spread and an isochron on the T-W plot, that seemingly was not permitted by the LA-ICP-MS data. Despite this the isochron overlaps within uncertainty with the majority of previously published LA-ICP-MS data for Jian-1 (Neymark et al., 2018), indicating that the accuracy of the measurements by micro-beam was relatively robust at the 1-3 % precision of single data points. However, the ID-TIMS lower intercept date of  $154.969 \pm 0.082$  Ma is ~1 % younger than the LA-ICP-MS weighted mean  $^{206}\text{Pb}/^{238}\text{U}$  date of  $156.55 \pm 0.36$  Ma ( $n = 40$ ; MSWD = 1.4) (Neymark et al., 2018). This offset would likely reduce further if a fractionation factor was renormalised to the ~0.4 % younger ID-TIMS intercept date of SPG-IV than was assumed by Neymark et al., (2018). However, this would still not fully account for the offset on the weighted mean at the  $\pm 0.23$  % 2s precision stated for Jian-1 (Neymark et al., 2018).



375 **Figure 45.** T-W plots of Jian-1. A) All ID-TIMS U-Pb data and isochron; B) ID-TIMS data excluding  
lowest Pb\*/Pbc and isochron lower intercept compared with LA-ICP-MS data for Jian-1 Cassiterite  
(Neymark et al., 2018).

## 7. Evaluating the accuracy of cassiterite U-Pb dating: The geochronology of Cligga Head, SW 380 England.

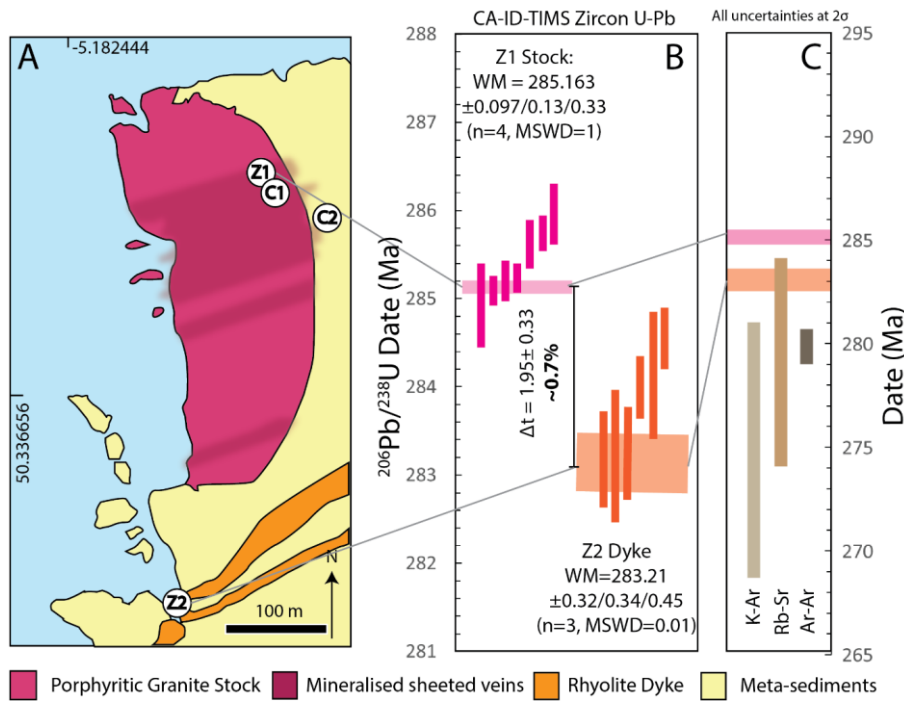
### 7.1 Geology and previous geochronology of Cligga head W-Sn deposit

The Cligga Head W-Sn greisen deposit is located at the central-northern periphery of the post-Variscan,  
Early Permian, SW England (Cornubian) Batholith, UK. Key features of the local geology (Fig. 56;  
Hall, 1971; Jackson et al., 1977; Moore and Jackson, 1977) are a porphyritic granite stock that intruded  
385 and locally thermally metamorphosed Devonian pelitic and psammitic meta-sediments at the contacts  
at about 1 kbar (Hall, 1971). The westerly extent of the granite has been eroded by the coast line. The  
main granite stock is cross-cut by a complex of sheeted quartz-muscovite (<1 % topaz and fluorite)  
greisen-bordered veins that predominantly contain quartz, in addition to tourmaline, chlorite, fluorite,  
cassiterite-stannite-arsenopyrite, wolframite and minor Cu sulfides. The cassiterite bearing greisen  
390 bordered veins extend across the contact into the meta-sediments.

Rhyolite porphyry dykes intruded ~150 m south of the stock. The absence of W-Sn mineralisation and  
greisenisation of the dykes, in addition to reported xenoliths of the granite greisen within the dykes  
(Reid and Scrivenor, 1906), indicates they occurred after the W-Sn mineralisation, although they  
contain disseminated chalcopyrite and supergene derivatives (Moore and Jackson, 1977).

395 Previous geochronology of muscovite from the greisenized zones indicated a date of ~280 Ma for the  
magmatic-hydrothermal system (Fig. 56). Two LA-ICP-MS U-Pb ages were presented (Moscati and  
Neymark, 2019) for cassiterite from Cligga Head. A TW lower intercept age of  $287.9 \pm 2.6$  Ma (n=59;  
MSWD = 2.2; initial  $^{207}\text{Pb}/^{206}\text{Pb}$  value of  $0.790 \pm 0.034$ ) or an anchored isochron with a lower intercept  
of  $289.2 \pm 2.5$  Ma (MSWD = 2.4). The weighted average Pbc corrected  $^{206}\text{Pb}/^{238}\text{U}$  date is  $288.4 \pm 3.0$   
400 Ma (n = 56, MSWD = 1.3). A second sample yielded lower intercept dates of  $287.2 \pm 4.8$  Ma (n=40;

MSWD = 2.1; initial  $^{207}\text{Pb}/^{206}\text{Pb}$  value of  $0.839 \pm 0.013$  and a Pbc corrected weighted average  $^{206}\text{Pb}/^{238}\text{U}$  date of  $285.2 \pm 4.2$  Ma ( $n = 38$ , MSWD = 1.3) (see Fig. 9).



**Figure 56.** A) Geological map of Cligga Head (after Moore and Jackson, 1977) with localities of zircon samples (Z1 and Z2) and cassiterite samples (C1 and C2) marked, pale blue signifies sea level; B) Zircon CA-ID-TIMS U-Pb dates and weighted mean ages for Cligga Head magmatism of the Porphyritic granite stock (Z1) and Rhyolite porphyry dykes. Dark shaded boxes represent single analyses of zircon tips, pale boxes represent weighted mean dates. The intrusions define a period of  $\sim 0.7\%$  of the absolute age in which the cassiterite may have formed. Uncertainties presented as  $\pm x/y/z$  where  $x$  = analytical uncertainties only for comparison with cassiterite ID-TIMS U-Pb dates also using the ET535 tracer;  $y$  = analytical and tracer calibration uncertainties;  $z$  = total uncertainty including  $^{238}\text{U}$  decay constant for comparison with ages derived from other decay systems. C) Comparison of zircon CA-ID-TIMS U-Pb weighted mean dates with previous geochronology for greisen system (Chen et al., 1993; Halliday, 1980).

415

## 7.2 Samples and Zircon CA-ID-TIMS U-Pb geochronology methods

Samples of the Granite Porphyry stock and the Rhyolite Porphyry dyke (Fig. 56) were analysed by zircon chemical abrasion (CA)-ID-TIMS U-Pb in order to provide constraints on the timing of cassiterite mineralisation. Zircon was analysed with the ET535 tracer and methods followed that of Tapster et al. (2016) using the same decay constants and  $^{238}\text{U}/^{235}\text{U}$  as the cassiterite analyses. Data is corrected for initial  $^{230}\text{Th}$  disequilibrium using a Th/U(melt) of 1.43 derived from a mean of granitic magmatism in the Cornubian Batholith (Simons et al., 2016). All data is contained within the supplementary material.

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### 7.3 Zircon CA-ID-TIMS U-Pb results and the “age-window” for Cligga Head cassiterite

The 10 analyses of zircon tip fragments from the host granite stock, yield a range in  $^{206}\text{Pb}/^{238}\text{U}$  CA-ID-TIMS dates from ~290 Ma to ~285 Ma. The youngest four dates form a statistically acceptable weighted mean of  $285.163 \pm 0.097$  Ma (MSWD = 1.01; n=4), which is interpreted as the emplacement age (Fig. 56).

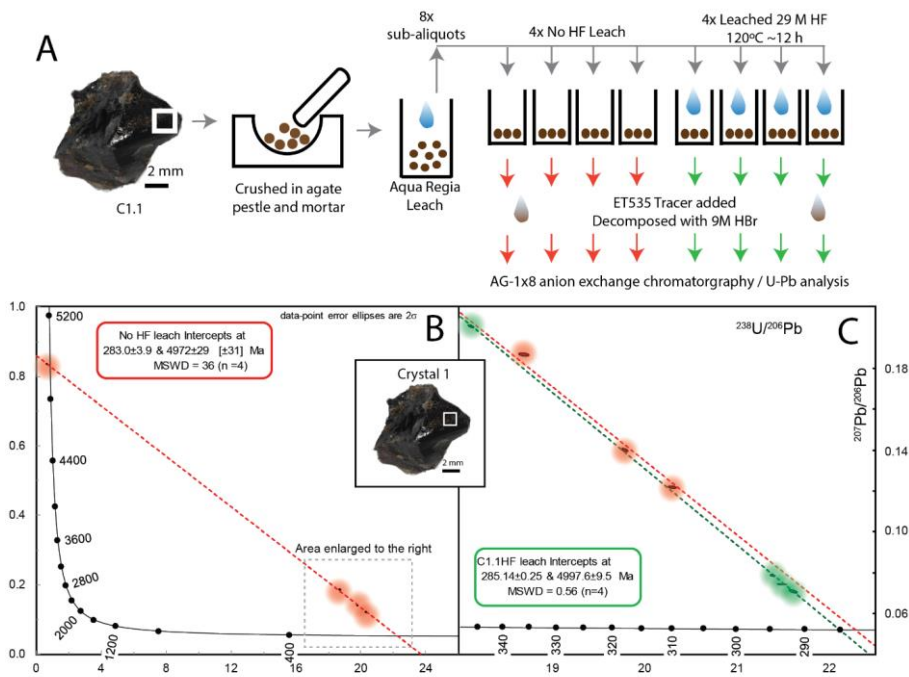
The 8 analyses of zircon tip fragments from the porphyritic dyke which intrudes to the SE of the stock, yield  $^{206}\text{Pb}/^{238}\text{U}$  dates from ~290 Ma to ~282 Ma. The youngest date, whilst overlapping with Concordia within its relatively low single data point uncertainty demonstrates high discordancy (~11 %) and does not form a single population weighted mean with any other dates, it is therefore attributed to residual Pb loss and rejected from the age interpretation. The youngest dates that form a statistically acceptable weighted mean date yield an age of  $283.21 \pm 0.032$  Ma (MSWD = 0.01; n=3) taken as the timing of emplacement (Fig. 56).

These dates for ~~the magmatism-magmatic events~~ at Cligga Head provide ~~independent~~ the minimum and maximum constraints on the absolute timing of the magmatic-hydrothermal greisen system ~~and the true “age” of cassiterite that is independent in which the cassiterite formed when taken in the context of geological constraints of the cassiterite isotopic dates.~~ They define a possible window for the “true” age of cassiterite of  $1.95 \pm 0.33$  Myrs, or about 0.7 % of the ~285 Ma absolute age.

### 7.4 Cligga Head Cassiterite ID-TIMS U-Pb geochronology

445 Cassiterite was sampled from two locations at Cligga Head (Fig. 56) from the first locality within the granite stock a single crystal (C1) was used within the decomposition experiments. In the first experiment a single part of the crystal was powdered into a single “parental” aliquot and divided into 8 approximately equal parts. We evaluated the effect of not leaching 4 of these cassiterite fractions in 29M HF relative to leaching, as described in the methods above and used in all of the other experiments (Fig. 67).

450 We then evaluated intra-crystal variation in U-Pb systematics by analysing an additional 3, spatially independent, parts of the same C1 crystal. To evaluate the inter-sample variation, fragments from 4 independent crystals were analysed from a second vein (C2) containing multiple cassiterite crystals that was sampled from within the meta-sediments. Cassiterite ID-TIMS methods followed those described above (Sect. 5). All cassiterite U-Pb data is contained within the supplementary material.



455 **Figure 67.** A) Schematic work flow of sampling and acid pre-treatment of C1.1 cassiterite. B) T-W plot and regression of non-HF leached sub-aliquots, individual data points are highlighted by a coloured

halo due to their size at the scale of spread in data. C) T-W plot and regression of C1.1 HF leached sub-  
460 aliquots (green) compared to the data and regression of non-HF leached (Red) sub-aliquots. Data are highlighted by a coloured halo due to their size at the scale of spread in data.

#### 7.4.1 Comparison of HF leached and non-HF leached cassiterite from the same parental sample

The sub-aliquots from the same parental powdered fragment of crystal (C1.1) that did not undergo HF  
leaching ranged from ~36 to ~55 pg of Pbc ( $^{206}\text{Pb}/^{204}\text{Pb} = 108$  to 207) and one aliquot yielded 7 ng of  
465 Pbc. Their regression line on the T-W plot forms a lower intercept of  $283.0 \pm 3.9$  Ma with a statistically unacceptable MSWD of 36. The four sub-aliquots that were HF leached contained 6.7 to 47 pg of Pbc ( $^{206}\text{Pb}/^{204}\text{Pb} = 97.7$  to 691). ~~Together they and yielded~~ a lower intercept age of  $285.143 \pm 0.25$  Ma with a statistically acceptable MSWD of 0.55 (n=4) and ~~a an upper y-~~intercept of  $0.8760 \pm 0.0058$ . None of the non-HF treated analyses form a single population (statistically acceptable MSWD) when  
470 combined with the regression line generated by the four HF treated sub-aliquots.

#### 7.4.2 Intra-grain cassiterite U-Pb variability

The three spatially independent aliquots from the same crystal (C1.2-C1.4) did not yield a statistically acceptable MSWD on their regression (lower intercept age =  $284 \pm 20$  Ma; MSWD = 7.2; n=3). Only  
475 one of the three spatially distinct fragments of the same crystal fractions forms a single population with the HF leached C1.1 regression line, forming a lower intercept age of  $285.19 \pm 0.22$  Ma (n = 5; MSWD = 0.73). All HF leached aliquots from the same crystal (C1) form a lower intercept age of  $285.67 \pm 0.72$  Ma with a statistically unacceptable MSWD of 10.1 (n = 7). The total Pbc amounts for these analyses ranges from ~3 pg to 8 pg and their  $^{206}\text{Pb}/^{204}\text{Pb}$  range from ~400 to ~620.

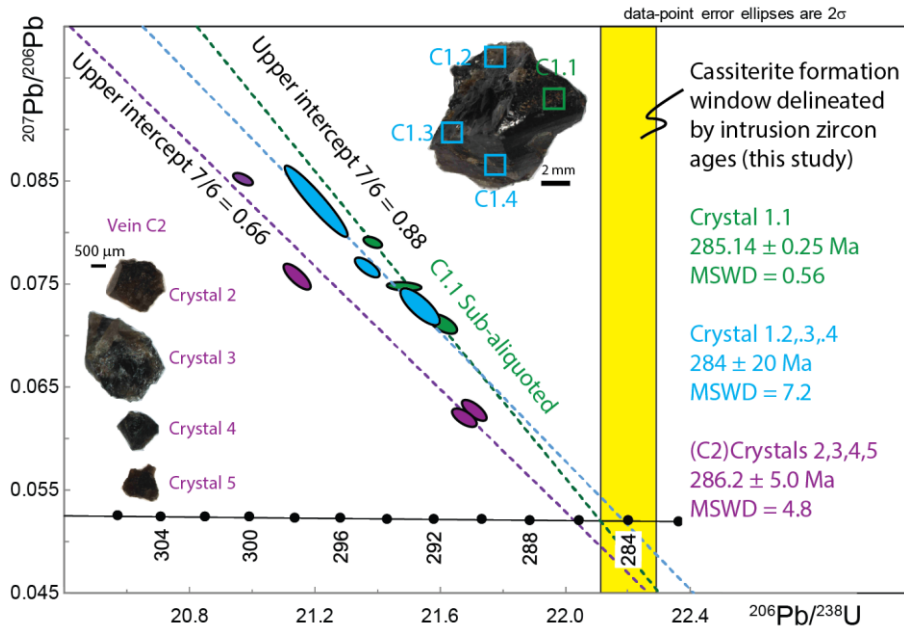
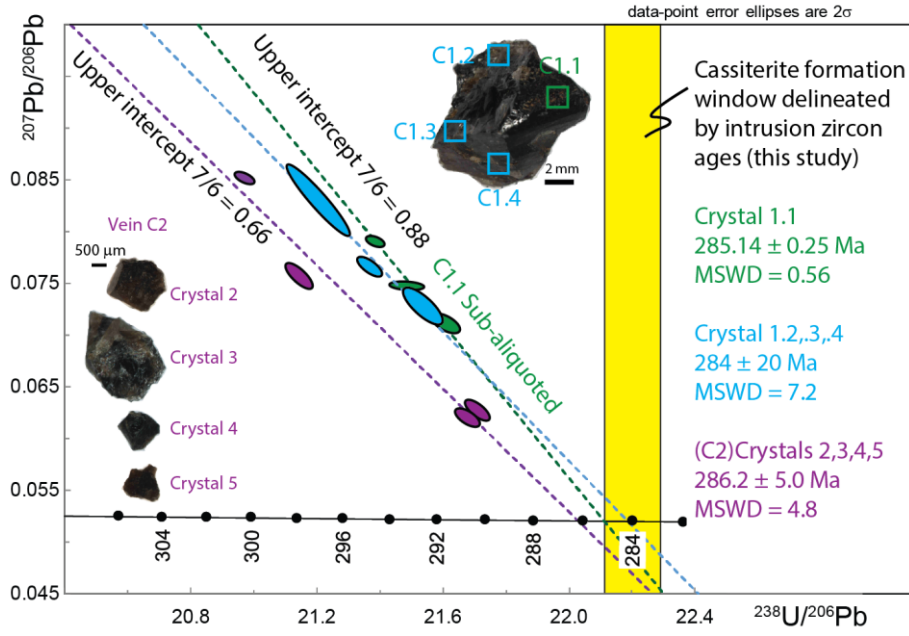
Anchoring the lower intercept to the previously determined ~285 Ma lower intercept date of C1.1, and  
480 thus estimating the minimum possible  $^{207}\text{Pb}/^{206}\text{Pb}$  ~~upper~~-intercept value due to the constraints offered by the granite age, yields  $^{207}\text{Pb}/^{206}\text{Pb}$  ~~upper~~-intercept of 0.848, 0.807 and 0.783 ( $\pm \sim 2\%$ ), lower than the previously determined ~~upper~~-intercept of ~0.876 for the same crystal.

#### 7.4.3 Inter-vein and -grain variation in U-Pb systematics

No analyses derived from individual crystals of the second hand sample (C2) form a single population when integrated with data that formed the isochron of C1.1 (Fig. 8).-Neither do the four analyses form a single population between themselves, yielding a lower intercept of  $286.2 \pm 5.0$  Ma with a statistically unacceptable MSWD = 48 (n = 4) and  $^{207}\text{Pb}/^{206}\text{Pb}$  ~~upper~~-intercept of 0.72 when freely regressed and 0.66 ( $\pm 5\%$ ) when anchored to the C1.1 lower intercept date and maximum possible age derived from the granite of  $\sim 285.1$  Ma (Fig. 8). The total Pb amounts for these analyses ranges from  $\sim 4$  pg to 17 pg and their  $^{206}\text{Pb}/^{204}\text{Pb}$  range from  $\sim 430$  to  $\sim 1300$ .

Again, to estimate the maximum initial  $^{207}\text{Pb}/^{206}\text{Pb}$  by anchoring the lower intercept of each date the maximum possible age intercept defined by the granite yields ~~upper-y-~~intercept initial  $^{207}\text{Pb}/^{206}\text{Pb}$  for each data point of 0.695, 0.643, 0.595 and 0.561 ( $\pm 0.7$ - $2.3\%$  2s), significantly lower than the value derived for C1.1 ( $\sim 0.876$ ).

495





500 **Figure 78.** T-W plot and regression lines for HF leached cassiterite from Cligga Head relative to possible age defined by zircon (Fig. 56). Regressions are for: C1.1 sub-aliquots (n=4) as presented in Fig. 67; Spatially independent samples of the same crystal (n=3); and independent crystals from a different vein (C2) (n=4), note ~~upper-y~~-intercept value is for regression anchored to maximum possible lower intercept as defined by the granite, it therefore represents a maximum ~~y-upper~~-intercept value. Only the sub-aliquots of C1.1 yield a regression without over dispersion. See text for further discussion.

## 505 **8. Discussion**

This cassiterite U-Pb dataset, combined with prior work (e.g. Li et al., 2016; Moscati and Neymark, 2019; Neymark et al., 2018; Zhang et al., 2017) highlights the potential for U-Pb dating of cassiterite using both microbeam and isotope dilution methods. Such datasets also reveal some complexities and issues that need to be considered when deriving meaningful ages from cassiterite dates.

### 510 **8.1 Assessing U-Pb systematics of (Cligga Head) cassiterite**

The range of ID-TIMS experiments described above permit us to evaluate the U-Pb systematics of cassiterite at high-precision and within the context of independent geochronological constraints derived from zircon CA-ID-TIMS dating within a well constrained geological model. There are three key features of the data set:

515 **1.** The lower intercept date formed by a single population of HF leached cassiterite sub-aliquots of the same parental material (C1.1), indicating binary mixing line between an initial Pb and a radiogenic Pb source. The lower intercept yields a precise date that is consistent with the indirect age constraints on cassiterite U/Pb\*. The effects of any initial isotopic disequilibrium have to be considered when deriving a sample age and uncertainty (see below).

520 **2.** The over-dispersion of non-HF leached cassiterite compared to HF leached C1.1 cassiterite analyses (Fig. 67) indicating non-binary mixing between initial and radiogenic Pb. The non-HF leached experiments contained a greater contribution of Pbc, with the high-MSWD indicating different sources/isotopic compositions of Pbc that varied in magnitude of contribution between sub-aliquots. Both sets of experiments (Fig. 67) were treated with aqua regia prior to analysis as with previous

525 cassiterite ID-TIMS studies (Gulson and Jones, 1992). These results therefore indicate that leaching  
with concentrated HF is an effective means to remove the Pbc contained within inclusions within the  
cassiterite that are exposed to the surface during crushing in a pestle and mortar. The results also indicate  
HF leaching is effective in removing Pbc bearing contaminants introduced by crushing in pestle and  
mortar, as proposed as a potential issue by Neymark et al. (2018). The acid resistant nature of cassiterite  
530 and ability of HF to dissolve silicate minerals without partially dissolving the cassiterite and potentially  
leading to incongruous removal of Pb or U, makes this an important step to remove non-lattice bound  
Pbc prior to decomposition by HBr and improve the accuracy of resulting ID-TIMS U-Pb dates.

3. The differences between spatially independent samples from within and between crystals, and  
between cassiterite from different veins (Fig. 78), indicating localised controls on the U-Pb systematics  
535 of cassiterite. Only one spatially independent fragment from the same crystal of cassiterite forms a  
single population with the isochron of HF leached C1.1 described above. A dataset based upon aliquots  
of the same crystals (C1.2, C1.3, C1.4) yield a similar age to the C1.1 isochron but is over-dispersed  
and the lower intercept is less precise. A similar cassiterite U-Pb ID-TIMS dataset from a different vein  
(C2) is also over-dispersed and plots distinctly to the left of the C1 mixing array between the constrained  
540 crystallisation age and initial Pbc  $^{207}\text{Pb}/^{206}\text{Pb}$  compositions.

Open system behaviour with respect to Pb (Pb-loss) would shift data points to the right (higher U/Pb)  
of the C1.1 (or similar) isochron. Estimated uranium concentrations (see Cassiterite ID-TIMS U-Pb  
data table in supplementary materials) are similar between cassiterite from the two veins and as such  
there is no apparent reason why uranium mobility would be manifest to a greater extent between  
545 different veins. We consider that the most likely explanation is that the over-dispersion results from  
variation in the initial Pbc of the cassiterite to more “radiogenic” initial  $^{207}\text{Pb}/^{206}\text{Pb}$  values than ~0.88  
as defined by the C1.1 isochron, potentially as low as ~0.56.

The majority of documented  $^{207}\text{Pb}/^{206}\text{Pb}$  from feldspars in granites and fluid inclusions in the Cornubian  
Batholith are ~0.8-0.9 (Hampton and Taylor, 1983; Wayne et al., 1996), with only one magmatic  
550 feldspar analysis yielding a  $^{207}\text{Pb}/^{206}\text{Pb}$  of ~0.55 (Hampton and Taylor 1983). It is unlikely that these  
low values reflect the primary input signature from the magmatic volatiles. Initial Pb with  $^{207}\text{Pb}/^{206}\text{Pb}$

this low are uncommon but are present in a global array of vein carbonates (Roberts et al., 2019). When the nature of cassiterite-bearing greisen systems, such as Cligga Head, is taken into account, the ability to generate atypical Pbc isotopic compositions is less surprising. Geisenisation at the margins of the cassiterite-bearing veins represents extensive, destructive, fluid-rock interaction and elemental exchange during alteration of host rocks by high temperature ~450-300°C, highly acidic, fluorine-rich (HF) fluids (Burt, 1981; Codeço et al., 2017; Lecumberri-Sanchez et al., 2017). As at Cligga Head, W-Sn greisen bordered vein magmatic-hydrothermal deposits are commonly hosted by, or are proximal to thick continentally derived sedimentary packages (Lecumberri-Sanchez et al., 2017) that likely represent the source of Sn enrichment prior to magma genesis (Romer and Kroner, 2016). These sediments will contain, older U- and Pb\*- rich continental mineral detritus (e.g. zircon, monazite) that can be extensively leached by the hot, F-rich, acidic ore-forming fluid generating significant spatial and temporal variations in the Pbc isotopic composition incorporated into the cassiterite lattice over the lifetime of single crystal precipitation, and the magmatic hydrothermal system as a whole. It is perhaps significant that the cassiterite that appears to contain the lowest initial Pbc component originates from a vein bordered by greisenized meta-sediment rather than hosted within the granitic stock.

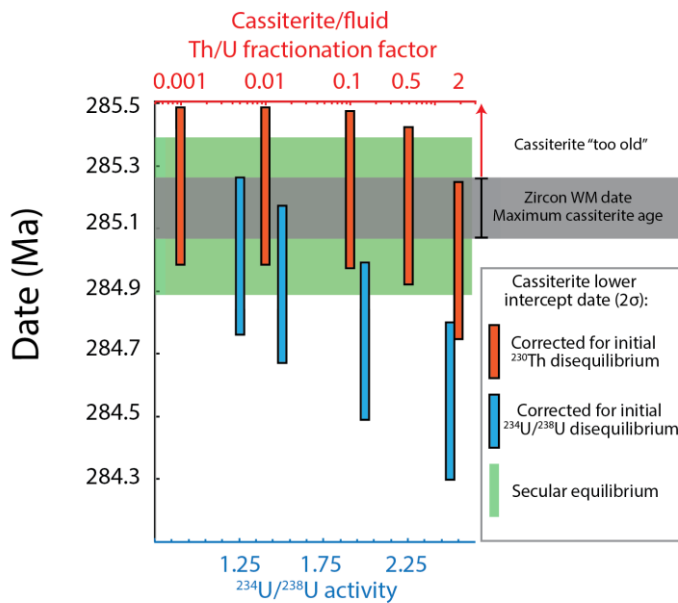
## 8.2 Cassiterite U-Pb ages and effects of initial U-Th disequilibrium.

The high-precision U-Pb lower intercept cassiterite date (C1.1; Figs. 6-7 and 78) comes with the caveat that it is not corrected for potential initial isotopic disequilibrium effects within the  $^{238}\text{U}$  decay chain. In zircon geochronology a deficit in  $^{206}\text{Pb}$  due to  $^{230}\text{Th}$  disequilibrium (Schärer, 1984) can be reasonably corrected using a reasonable assumption of the mineral/melt partition coefficient or an estimate of melt Th/U, and an estimate of mineral Th/U, typically based on assumed  $^{208}\text{Pb}/^{232}\text{Th}$ - $^{206}\text{Pb}$ - $^{238}\text{U}$  concordance for zircon ID-TIMS data. Cassiterite typically has extremely low  $^{232}\text{Th}$  relative to U (Th/U =  $<10^{-4}$  Neymark et al., 2018). This geochemical feature may be derived from preferential partitioning of U into the ore fluids, and/or preferential partitioning of U upon precipitation of cassiterite leading to a deviation from secular equilibrium. Both processes will require a significant correction for a deficiency in  $^{206}\text{Pb}$  due to a low  $^{230}\text{Th}/^{238}\text{U}$  activity in the initial cassiterite. Alternatively this geochemical feature could

580 simply be explained by processes that do not preclude secular equilibrium with regards to  $^{230}\text{Th}$  upon cassiterite crystallisation. These include low  $^{232}\text{Th}/^{238}\text{U}$  magmatic ore fluid sources, or the contribution of localised U-rich components to fluids that are in secular equilibrium with regards to  $^{230}\text{Th}$  and  $^{238}\text{U}$  during alteration around veins. In these cases the  $^{232}\text{Th}$ - $^{208}\text{Pb}$  system would not provide a realistic proxy for initial  $^{230}\text{Th}$  disequilibrium in cassiterite.

585 It must also be considered that the hydrothermal fluids, and therefore the initial cassiterite may not have been in secular equilibrium for  $^{238}\text{U}$ - $^{234}\text{U}$ . A relative excess of  $^{234}\text{U}$ , or a  $^{234}\text{U}/^{238}\text{U}$  activity ratio  $>1$ , at the time of cassiterite precipitation will result in an excess of  $^{206}\text{Pb}/^{238}\text{U}$ , an older measured date relative to the true age. These effects therefore counteract effects of  $^{206}\text{Pb}$  deficiency due to  $^{230}\text{Th}$  disequilibrium. Mixing with meteoric fluids appears to be an important process in cassiterite formation (Fekete et al., 2016) and as highlighted by the compilation of Roberts et al. (2019) the  $^{234}\text{U}/^{238}\text{U}$  activity of crustal 590 fluids can be in excess of 1, (shallow ground waters median value of 2.25; hydrothermal fluids median of 1.41).

Estimates of the effects of initial disequilibrium (Fig. 89) indicates that corrections for the exclusion (or absence) of initial  $^{230}\text{Th}$ , potentially inferred by the characteristic low  $^{232}\text{Th}/\text{U}$  in cassiterite, produces a maximum possible date ( $\sim 285.23$  Ma) that is still within uncertainty of the maximum age permitted by the granite host. Effects of excess initial  $^{234}\text{U}$  indicate that an activity ratio of 1.5 produces dates  $\sim 200$  595 kyrs younger (284.92 Ma) than that of secular equilibrium, yet also lies within the analytical uncertainties of the host granite. In the more extreme scenario of a  $^{234}\text{U}/^{238}\text{U}$  activity ratio of 2.5, this reduces to 284.55 Ma and would be significantly younger.



600 **Figure 89.** Effects of corrections for initial intermediate daughter disequilibrium on the lower intercept date of C1.1 regression over geologically feasible parameters. Note that corrections are treated independently and that combinations of initial disequilibrium effects could be present. The maximum possible age of cassiterite is defined by the zircon age and uncertainty of the Porphyritic Granite stock that hosts the cassiterite bearing veins. See text for discussion.

605

Not accounting for the uncertainty that arises from the potential effects of disequilibrium, the lower intercept of the four point isochron C1.1 yields a precision of  $\pm 250$  kyrs or 0.088 % ( $2\sigma$ , analytical uncertainties only,) and an agreement with the possible window for hydrothermal activity defined by the CA-ID-TIMS zircon dates of the intrusions ( $\sim 0.7$  %). Moreover it also demonstrates a clear temporal association with the granitic stock that hosts the Cligga Head deposit (the cassiterite date is  $0.021 \pm 0.268$  Myr younger than the zircon date for the granitic stock) (Figs. 89, 910). High precision constraints on the durations of relatively simple ore-forming magmatic-hydrothermal systems spatially associated with single intrusions indicates timeframes  $< 10$ s to 100s kyrs (Li et al., 2017; Tapster et al., 2016). It is therefore feasible that the emplacement of the host granite stock, cooling to the 400°C-350°C

615 temperature interval of cassiterite precipitation (Smith et al., 1996), and transfer of ore forming volatiles  
from depth, all occurred within the timeframes defined by the <0.1 % uncertainty of the zircon and  
cassiterite data presented here. However, considering the potential uncertainty due to initial U-Th  
disequilibrium effects over geologically reasonable assumptions the cassiterite uncertainty could be  
620 expanded by ca. -200 kyrs and would be asymmetric (Fig. 89). At present the poor understanding of  
isotopic disequilibrium effects due to elemental partitioning into ore fluids, and cassiterite represent a  
limitation for the accurate interpretation of absolute ages of cassiterite beyond the <~500 kyr timescale.

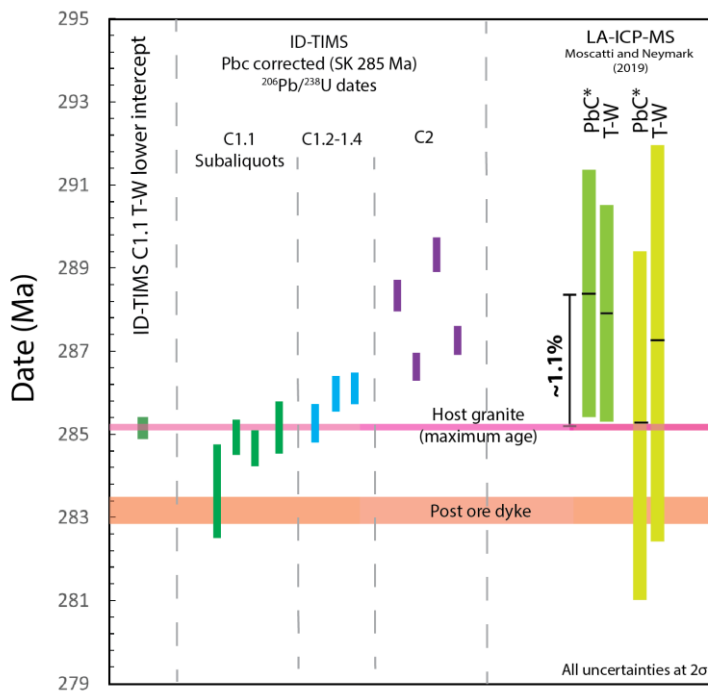
### 8.3 Implications and strategies for LA-ICP-MS U-Pb cassiterite geochronology

In this study we have generated ID-TIMS U-Pb data on cassiterite that have undergone full  
625 decomposition in a single stage. These include samples that have previously been used for LA-ICP-  
MS studies using an approach to U/Pb normalisation using an ad hoc cassiterite reference material with  
an inferred U/Pb age. The comparison of the ID-TIMS and LA-ICP-MS U-Pb data provide a direct  
means to assess the accuracy of the microbeam dates (Moscati and Neymark, 2019; Neymark et al.,  
2018), which show agreement at the ca. 1 % level and is therefore comparable to the quoted levels of  
630 accuracy and similar to the accuracy for U-Pb (zircon) studies (Horstwood et al., 2016). Additional  
characterisation and over a wider range of cassiterite materials by the ID-TIMS methods described here  
will establish a focal point for the refinement of cassiterite measurement uncertainties by LA-ICP-MS.

The ID-TIMS U-Pb data from the Cligga Head cassiterite strongly suggests that the initial Pb isotopic  
composition varies and that a simple binary mixture between a radiogenic and initial Pb may not be  
635 expected within and between crystals. A high proportion of LA-ICP-MS U-Pb cassiterite datasets  
record over-dispersion (e.g. ~2/3 of samples, typically analysed from single crystals, reported by  
Neymark et al., 2018) suggesting that these are also not strict binary mixtures. We postulate that some  
of the over-dispersion can be attributed to variable initial Pb isotopic compositions within cassiterite  
that is inherited from the hydrothermal systems. This complexity in the U-Pb systematics represents a  
640 potential limitation on the accuracy of the resulting age interpretation of cassiterite. This limitation is  
interdependent with the single data point precision of the U-Pb analyses and the ability to resolve

different populations of the Pbc mixing endmember that will bias the regression of a data set away from the accurate lower intercept.

We explore how the U-Pb data can be leveraged by variable Pbc by correcting the Cligga Head data set  
 645 by using a widely employed strategy for minerals that accommodate Pb into their lattice upon formation, of correcting the component of initial Pbc using a model Pbc value at ~285 Ma (Stacey and Kramers, 1975). Figure 9-10 demonstrates the resulting Pbc corrected  $^{206}\text{Pb}/^{238}\text{U}$  dates range from being in good agreement with the host granite age to being 'too old' by ca. 4 Myrs (~1.5 %). Without the maximum constraint from the age of the granite, these 'model' U-Pb cassiterite dates alone could be misinterpreted  
 650 as a protracted episode of mineralisation. Comparable shifts, both in magnitude and direction, can be observed (Fig. 10) within one of the two samples from the same deposit that were analysed by LA-ICP-MS U-Pb (Moscati and Neymark, 2019; Fig. 9).



655 **Figure 910.** The effects of using a correction for Pbc on the  $^{206}\text{Pb}/^{238}\text{U}$  dates of single analyses of cassiterite, in this case a model Pbc value at 285 Ma (Stacey and Kramers, 1975) was used. Data are shown relative to the freely regressed ID-TIMS lower intercept of C1.1 that shows excellent agreement with the host granite age. Pbc corrected dates range from being in good agreement with the age constraints to being offset by ~4 Myrs relative to the maximum possible age of cassiterite defined by zircon dates for the granite. The samples analysed in a previous LA-ICP-MS study from Cligga Head (Moscati and Neymark, 2019) illustrate similar magnitudes and direction of offset can be observed in 660 both the T-W regressions. Despite the agreement of the Pbc corrected (Pbc\*) weighted mean  $^{206}\text{Pb}/^{238}\text{U}$  date 290 Ma (Stacey and Kramers, 1975) of one sample, this approach to data interpretation is invalid, see text for discussion.

665 ~~To further~~ Figure 11. Illustrates how the ability to resolve this effect ~~this effect, and~~ and the interplay with the impact of larger ~~lower~~ precision on single data points can impact ~~uncertainties~~ the accuracy of the regression and age interpretation. ~~on resolving dispersion, an~~ The isochron we construct is based upon all Cligga Head cassiterite ID-TIMS U-Pb analyses, but with uncertainties expanded to  $\pm 1\%$  ( $2\sigma$ ) precision. (Fig. 11). It shows no dispersion (MSWD = 1.2, ~~n=12 and statistically acceptable~~), however 670 the resulting age is biased too old (lower intercept at  $286.80 \pm 0.95$  Ma; ~~MSWD = 1.2, n = 12~~) for the known maximum possible age that is defined by the granite.

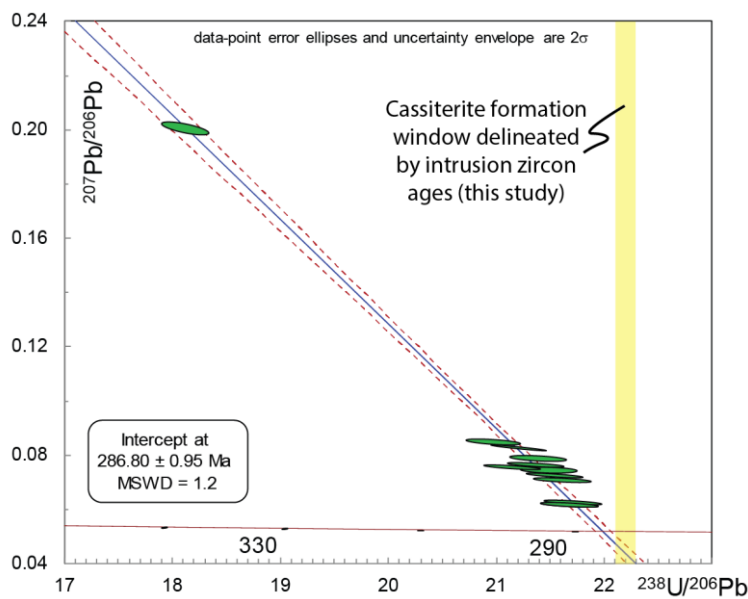
The approach of taking weighted means of common Pb corrected  $^{206}\text{Pb}/^{238}\text{U}$  data is strongly advised against, as it diminishes the evaluation of dispersion within the data and ignores the uncertainty of  $^{207}\text{Pb}/^{206}\text{Pb}$  measurement (see Horstwood et al., (2015,) for community discussion of this issue in LA-ICPMS U-Pb). However, to demonstrate this approach does not resolve the issues of ~~but as an illustration of the effects of~~ variable initial Pbc composition, the weighted mean date of Pbc corrected  $^{206}\text{Pb}/^{238}\text{U}$  dates for the same data set with uncertainties expanded to 1% is  $286.23 \pm 0.83$  Ma (MSWD = 1.3; ~~n = 12~~) for the same data set with uncertainties expanded to 1% and therefore despite its seemingly precise date, it also lies outside of uncertainty of the maximum possible age defined by the granite 675 ~~weighted mean date of Pbc corrected~~  $^{206}\text{Pb}/^{238}\text{U}$  dates  $286.23 \pm 0.83$  Ma (MSWD = 1.3; ~~n = 12~~). 680

Any bias resulting from Pbc variation would be in addition to greater scatter on individual data points resulting from lower precision measurements. This reinforces the limitations on the accuracy of the



interpreted ages when derived from high-n but lower precision data sets ~~when there are potential variations in the initial Pb isotopic composition, because they can due to heterogeneity being be masked~~ by the single data point uncertainty ~~yet impact on the accuracy of the age.~~ In addition to the limits of systematic uncertainties defined by validation RMs, ~~this limitation~~ should be taken into account, ~~in addition to the limits of systematic uncertainties defined by validation RMs,~~ within future micro-beam work of both cassiterite, ~~alongside any and other minerals phase that contains with initial Pb.~~

We note that the reference values of any cassiterite material used as an RM should not be corrected for Pb, or any initial isotopic disequilibrium effects, rather they should reflect the isotopic values of material ablated (Horstwood et al., 2016), as with the data presented in Figs. 34, 45, 6-7 and 78. As HF leaching of cassiterite targets inclusions, ~~it and~~ is unlikely to affect lattice bound Pb ~~we suggest this will not and~~ modify the isotope ratios as chemical abrasion does for zircon RMs.



**Figure 1011.** Illustration of how variable Pb compositions in cassiterite form a “single population”, yet with an inaccurate T-W lower intercept within lower single data point precision data sets. The data are all HF leached ID-TIMS U-Pb data from Cligga Head with artificially expanded single data point uncertainties of 1% (2s) for both  $^{238}\text{U}/^{206}\text{Pb}$  and  $^{207}\text{Pb}/^{206}\text{Pb}$ . This level of uncertainty is therefore a

700 reasonable illustration of the upper range of analytical precision that may be achieved through  
microbeam techniques in relatively low U concentration materials such as cassiterite. Note the data with  
artificial uncertainties does not show excess dispersion (statistically acceptable MSWD) but produces  
a relatively precise lower intercept (0.33 %) outside of uncertainty and older than the maximum  
705 permitted age of cassiterite delineated by the granite (Fig. 56). This does not factor in any additional  
systematic or analytical inaccuracy from lower precision methods.

## 9. Conclusions

The absolute dating of Cassiterite using the U-Pb decay system offers the potential to link the  
hydrothermal processes responsible for Sn mineralisation to the regional and local (deposit) scale  
710 magmatic-hydrothermal systems that may or may not be related to their genesis. Recent studies (e.g.  
Liu et al., 2007; Moscati and Neymark, 2019; Neymark et al., 2018; Yuan et al., 2011, 2008; Zhang et  
al., 2017; and others) have demonstrated the potential of cassiterite for U-Pb geochronology using  
microbeam methods (LA-ICP-MS) applied to the mineral from a wide range of deposits. The lack of  
materials that have characterised U-Pb compositions via total dissolution isotope dilution methods, and  
715 the difficulties in obtaining these, has led to the development of calibrations using inferred ages for ad  
hoc reference materials and impedes the verification of U-Pb datasets generated.

We have demonstrated that useful amounts of cassiterite can be fully decomposed under closed system  
conditions, using readily obtainable and low Pbc blank HBr acid, over timescales and using apparatus  
similar to that used in zircon ID-TIMS work. Increasing the surface area/volume ratio (e.g., using a  
720 pestle and mortar) is in fact an advantage when utilised with an HF leaching step, as it not only expedites  
cassiterite decomposition, but will also expose Pbc-bearing inclusions that can be leached and rinsed  
away, along with any material abraded during crushing in a pestle and mortar. The notorious acid  
resistance of cassiterite ~~here-~~works to our advantage as HF leaching leaves the lattice bound U-Pb  
systematics intact. The methodology presented here indicates that Pb can be isolated using a simple  
725 modification of AG-1x8 resin HCl based anion exchange chromatography typically used for zircon,  
with no detectable penalties for ionisation by TIMS, and a further AG-1x8 HCl and HNO<sub>3</sub> based scheme  
is effective for the U elution.

The U-Pb ID-TIMS data presented for a range of materials between ~150 to ~1600 Ma, demonstrate that cassiterite ID-TIMS U-Pb geochronology can potentially generate U-Pb lower intercept dates at ±  
730 <0.1 % precision, from a relatively low number of analyses. Using the classic example of the W-Sn Greisen deposit at Cligga Head, UK, we demonstrate that the implementation of “internal isochrones” from sub-aliquots of powdered fragments can ~~am~~ derive a U-Pb ID-TIMS cassiterite age interpretation consistent with an independent high-precision zircon CA-ID-TIMS U-Pb constraint on the age of cassiterite. These direct constraints show that the hydrothermal system was relatively short lived  
735 illustrating the potential of this method to define causative magmatic events. However, unknowns regarding initial isotopic disequilibrium of cassiterite probably limit the true uncertainty on the age to ~400 kyrs.

Comparison of ID-TIMS and previously reported LA-ICP-MS U-Pb datasets for the three samples indicate that the latter are accurate at the ~1 % ( $2\sigma$ ) level, especially when taking into account the ID-  
740 TIMS date that was ~0.4 % younger than the inferred age for the SPG-IV cassiterite used to normalise matrix effects in the Neymark et al. (2018) study. This direct U/Pb determination can be used to refine LA-ICP-MS data that are normalised to this ad hoc reference material, further improving the underpinning of cassiterite U-Pb via LA-ICP-MS. Further total digestion ID-TIMS U-Pb analyses of  
745 cassiterite will further improve the method, hopefully providing additional reference materials for calibration and verification.

The high precision data presented here indicates that intra- and inter- grain variations occur in the isotopic composition of Pbc present within the cassiterite. The nature of greisen Sn hydrothermal systems and their association with continentally derived sedimentary packages, may be conducive to  
750 localised and variably radiogenic Pbc compositions. The potential for these variations in Pbc compositions places limitations on the interpretive power of data sets as a function of single data point precision. A cautious approach to the age interpretation of lower precision micro-beam data sets, beyond the issues of analytical precision, is advised.

755 **10. Data availability**

Data to reproduce all plots presented here is provided within the supplementary material excel workbook.

**11. Author contributions**

760 ST and JB designed the experiments that were carried out by JB with contributions from ST. ST interpreted the data and prepared the manuscript with contribution from JB.

**12. Competing interests**

The authors declare no competing interests.

**12. Acknowledgements**

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