

Short communication: Synchrotron-based elemental mapping of single grains to investigate variable infrared-radiofluorescence emissions for luminescence dating

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Abstract. During ionizing irradiation, potassium (K)-rich feldspar grains emit infrared (IR) light, which is used for infrared-radiofluorescence (IR-RF) dating. The late-saturating IR-RF emission centred at ~880 nm represents a promising tool to date Quaternary sediments. In the present work, we report the presence of individual grains in the K-feldspar density fraction displaying an aberrant IR-RF signal shape, whose combined intensity contaminates the sum signal of an aliquot composed of dozens of grains. Our experiments were carried out at the National Synchrotron Light Source (NSLS-II) at the submicron resolution X-ray spectroscopy (SRX) beamline. We analysed coarse (> 90 μm) K-feldspar bearing grains of five samples of different ages and origin in order to characterize the composition of grains yielding the desired or contaminated IR-RF emission. Using micro-X-ray-fluorescence ($\mu\text{-XRF}$), we successfully acquired element distribution maps of up to 15 elements (<1 μm resolution) of sections of full grains previously used for luminescence dating. In keeping with current theories of IR-RF signal production, we observed a trend between the relative proportions of Pb and Fe and the shape of the luminescence signal: most grains with the desired IR-RF signal shape had high Pb and low Fe contents. Interestingly, these grains were also defined by high Ba and low Ca contents. Our study also represents a proof-of-concept for mapping the oxidation states of Fe using micro-X-ray absorption near-edge structure spectroscopy ($\mu\text{-XANES}$) on individual grains. The high spatial resolution enabled by synchrotron X-ray spectroscopy makes it a powerful tool for future experiments to elucidate long-standing issues concerning the nature and type of defect(s) associated with the main dosimetric trap in feldspar.

33 1 Introduction

34 Geochronologic data provide essential information for understanding the rates of Earth's surface processes, environmental
35 changes, and the evolution of life. Advances in dating techniques have fundamentally changed our capacity to piece together
36 our evolutionary past over millions of years, with luminescence dating proving to be a powerful tool in this field as it applies
37 to various types of sediments and contexts. The technique determines an age estimate for when mineral grains were last
38 exposed to daylight or heat. Luminescence dating methods rely on the property of certain minerals to record the amount of
39 radiation to which they have been exposed during burial and release energy when exposed to sunlight or high temperature
40 (e.g., Aitken, 1985, 1998; Bateman, 2019). In the laboratory, the total amount of energy per unit mass stored in the mineral is
41 measured (dose, Gy). The energy absorption rate per unit mass (dose rate, Gy a^{-1}) is derived from knowledge of the natural
42 radioactivity surrounding the sampled sediment. The quotient of these two values (dose/dose rate) gives the burial age.

43 Of the two minerals routinely used for luminescence dating of sediments, quartz and potassium (K-) rich feldspar, the
44 latter allows for the dating of older deposits of up to ~600 000 years (considering a dose rate of 1 Gy ka^{-1}) using infrared
45 stimulated luminescence (IRSL, Hütt et al., 1988). The datable upper age limit is given by the IRSL signal saturation after
46 exposure to radiation doses around 600 Gy (see summary in Sec. 8.1 in Murari et al., 2021a). Over the past decades, different
47 methods have been proposed to extend this upper age limit with varying degrees of success. The infrared-radiofluorescence
48 (IR-RF) signal of K-feldspar is a promising candidate for such an extension. The RF signal arises from prompt radiative
49 recombination of charge within crystalline materials during continuous exposure to ionizing radiation. The IR-RF emission at
50 880 nm (e.g., Kumar et al., 2018; Riedesel et al., 2021; Sontag-González et al., 2022) decreases in intensity with dose
51 accumulation as the electron traps fill until saturation (Trautmann et al., 1999a). This saturation level constrains the time range
52 over which IR-RF dating is applicable.

53 Murari et al. (2018) demonstrated that an accurate dose recovery of a known dose of 3600 Gy is possible (a dose
54 recovery test is a laboratory performance check of the measurement protocol, and successful dose recovery is a prerequisite
55 for any protocol). If we assume typical environmental dose rates of between 3 Gy ka^{-1} and 1 Gy ka^{-1} , then IR-RF dating could
56 produce age estimates ranging from 1.2 Ma to 3.6 Ma, which is around four times greater than the upper dating limit of
57 conventional luminescence dating methods. However, more recent studies (Murari et al., 2021b; Kreutzer et al., 2022)
58 indicated a dose saturation at around 1500 Gy, reducing the previously assumed temporal limit of IR-RF dating. Hence, the
59 uncertainty surrounding its upper age limit remains and further studies on known-age samples are required to assess whether
60 the sample/grain geochemistry influences the age limit. There is undoubtedly a gap in our current understanding of the
61 luminescence production processes in K-feldspar, and a revised conceptual model might be needed.

62 The vast majority of IR-RF studies have been performed on multi-grain aliquots, so the possible effects of variability
63 of the IR-RF signal from different grains has not received much attention in the literature so far, as detailed in section 2. Here,
64 we investigate the IR-RF signal of five samples from different locations at single-grain resolution and discuss what effect the
65 observed variability could have on multi-grain aliquots. To assess whether the grain geochemistry influences the IR-RF signal

66 and potentially the age limit of IR-RF, we examined individual K-feldspar grains at the submicron resolution X-ray
67 spectroscopy (SRX) beamline at the National Synchrotron Light Source II (NSLS-II) at Brookhaven National Laboratory.
68 Measurements at such a high-resolution may lead to a better understanding of the luminescence kinetics in feldspars. We report
69 on the feasibility and practicality of using μ -X-ray fluorescence (μ -XRF) and μ -X-ray absorption near-edge fine structure (μ -
70 XANES) techniques in investigating the luminescence signal origin and kinetics in K-feldspar.

71 **2 Method and rationale**

72 Identification of the defect type linked to the IR-RF signal and its concentration would enable us to better characterize the light
73 emission (signal sensitivity) in different types of feldspar, while identification of the origin of possible contamination in the
74 IR-RF signal could help us to gain a better understanding of the apparent early saturation or quenching of the IR-RF signal. μ -
75 XRF and μ -XANES produce high-resolution maps of elements and their oxidation states and are well suited for the purposes
76 of our study. μ -XRF elemental analyses are based on the characteristic fluorescence of atoms when stimulated with X-rays
77 with a higher energy than their ionization energy. In the case of μ -XANES, initial measurements of standards are run by
78 varying the incident beam energy to determine the specific energy equal to the absorption edge (binding energy of inner shell
79 electrons) of the element or ion of interest. This is apparent by an abrupt rise in the resulting fluorescence, which is different
80 between oxidation states as they require different minimum stimulation energies before ionization and subsequent
81 fluorescence. μ -XRF maps using the obtained absorption edge energies allow for maps of the different oxidation states of the
82 same element.

83 The use of synchrotron μ -XRF allows us to improve the spatial resolution compared with a standard lab-bench μ -
84 XRF setup (e.g., Buylaert et al., 2018) by reducing the beam spot size from $\sim 25 \mu\text{m}$ to $1 \mu\text{m}$ or $0.5 \mu\text{m}$. Though both the grain
85 geochemistry and crystallography should be investigated to characterize the defect type and its environment, in the present
86 study, we focus only on geochemistry.

87 The defect(s) responsible for the IR-RF emissions are still subject to debate. It has been suggested that IR-RF occurs
88 as a result of the change in the oxidation state of the participating lead (Pb) defect via the transition: $\text{Pb}^{2+} \rightarrow (\text{Pb}^+)^* \rightarrow \text{Pb}^+$
89 (Nagli and Dyachenko, 1986; Erfurt, 2003). A similar transition has been suggested for amazonite (see Ostrooumov, 2016),
90 but the direct connection between the Pb-centre and IR-RF has not yet been evidenced. Other reactions involving Pb^{4+} would
91 also be possible but haven't yet been formally proposed. Additionally, the IR-RF signal is composed of at least two separate
92 emissions. Previous publications placed the main IR emission at 1.43 eV (865 nm) based on Trautmann et al. (1999a, b) and
93 Erfurt and Krbetschek (2003), but more recent work including corrections for the spectrometer efficiency places the IR
94 emission closer to 880 nm (Kumar et al., 2018; Riedesel et al., 2021; Sontag-González et al., 2022). A second IR emission
95 centred at 955 nm (1.30 eV) at lower intensity has also been identified (Kumar et al., 2018), which partly overlaps with the
96 880 nm peak.

97 The presence of iron (Fe) in feldspar is known to lead to red RF (e.g., Telfer and Walker, 1978; Brooks et al., 2002;
98 Visocekas et al., 2014), with the maximum peak wavelength varying between 700 nm and 770 nm depending on feldspar
99 composition (Dütsch and Krbetschek, 1997; Krbetschek et al., 2002). Such observations are in line with the suggestion of more
100 than one component in the red photoluminescence of K-feldspar (Prasad and Jain, 2018). Despite the occurrence of the red RF
101 emission in Fe^{3+} state, its initial state remains a subject of many debates, with conflicting opinions suggesting either Fe^{2+} (Fe^{2+}
102 $+ h \rightarrow \text{Fe}^{3+}$; here h stands for hole) or Fe^{4+} ($\text{Fe}^{4+} + e^- \rightarrow \text{Fe}^{3+}$; here e^- stands for electron) (Kirsh and Townsend, 1988; Jain et
103 al., 2015). Recently, Kumar et al. (2020) argued that the initial state must be Fe^{4+} based on their findings using
104 cathodoluminescence microscopy. Spectral analyses showed that, with dose exposure, the red RF emission (~710 nm emission
105 in K-feldspar) increases, while the 880 nm emission decreases (Krbetschek et al., 2000; Erfurt and Krbetschek, 2003; Kumar
106 et al., 2018; Frouin et al., 2019). The thermal stability of the ~710 nm emission has been, however, questioned (e.g., Krbetschek
107 et al., 2000). Such a reduced thermal stability might be an issue for IR-RF dating, as it has been suggested that the tail of the
108 ~710 nm emission overlaps with the 880 nm emission, thus potentially playing a role in the shape of the measured IR-RF.
109 Such a contribution can be reduced to less than 5% of the IR-RF signal by using a bandpass filter centred at 850 nm (FWHM
110 40 nm), but can still affect the equivalent dose (D_e) value (see Sontag-González and Fuchs, 2022). D_e values are determined
111 by sliding the IR-RF dose-response curve of grains containing the natural signal onto that obtained after a full bleach of the
112 same aliquot. In summary, although previous studies have identified factors that may influence the IR-RF signal in several
113 ways, e.g., whether the IR-RF signal originates from Pb, and is affected by the presence of Fe^{2+} or Fe^{4+} , a conclusive
114 confirmation or comprehensive linkage between these factors is yet to be established.

115 A possible variability of the several RF emissions in individual grains has received scant attention, so far. Trautmann
116 et al. (2000) were the first to analyse the IR-RF signal of individual K-feldspar grains. Using spectral measurements of 21 to
117 42 grains from three samples, they observed up to four emissions (IR, red, yellow, blue) with variable intensities (a fourth
118 sample appears in their figure 3 but is not mentioned in the main text). An IR-RF dose-response curve was only reported for
119 one grain, which had a similar shape, albeit a later onset of saturation, when compared to the response from the multi-grain
120 aliquot of the same sample. More recently, Mittelstraß and Kreutzer (2021) analysed 60 grains from two samples, of which
121 55% and 80% emitted a detectable signal. In that study, between one and three grains per sample (~9% of signal-emitting
122 grains for both samples) were rejected due to a bad match between the natural and regenerative curves, which might have been
123 caused by equipment issues, but also due to sensitivity changes (Varma et al., 2013). However, all grains that emitted a
124 detectable signal displayed the expected decay shape for IR-RF (decreasing signal with increasing dose). Likewise, own
125 laboratory observations indicated that the signal varies in sensitivity across feldspar minerals and can be contaminated for
126 various reasons, leading to spectral interference or quenching, ultimately influencing the saturation level and/or the shape of
127 the IR-RF signal (Frouin et al., 2017, 2019; Kumar et al., 2020).

128 To investigate these issues, first, we recorded IR-RF curves from individual grains in our luminescence dating
129 laboratory at the Research Laboratory for Archaeology and the History of Art (RLAHA) at the University of Oxford (UK).
130 Then, during our beam time (96 h), and as a proof of concept, we optimized the measurement conditions and obtained

131 compositional maps of the individual K-feldspar grains. We paid particular attention to K, Ca, Fe and Pb. After analysing the
132 μ -XRF maps, μ -XANES measurements were done at selective spots where Fe and Pb occurred in greater concentrations. Note:
133 The atomic number of sodium (Na; another end member of the feldspar ternary system) is too low to be measured at the current
134 SRX beamline.

135 **3 Material and instrumentation**

136 A total of five samples were selected to represent a diversity of i) geological context, ii) geochemistry, iii) shape of the IR-RF
137 signal, and iv) age. Sample Gi326 from a Triassic sandstone from Bayreuth, Germany, is composed of 89% of K-feldspar
138 (Sontag-González and Fuchs, 2022) and has previously been used as a reference sample in a laboratory comparison of IR-RF
139 dating (Murari et al., 2021b). X7343 was collected from a Pliocene sediment at the Nyayanga site in the Homa peninsula,
140 Kenya (Plummer et al., 2023). X7363 was taken from the Gele Tuff in the Turkana Basin, Kenya, and dated by Ar/Ar at 1.32
141 ± 0.2 Ma (Phillips et al., 2023). X7368 is a sediment sample collected above the Silbo Tuff (0.751 ± 0.022 Ma, McDougall
142 and Brown, 2006) and below the Kale Tuff (younger than the Silbo Tuff but not directly dated) in the Turkana Basin, Kenya.
143 Sample H22550 is a coastal marine sample from Sula, Russia, dated by quartz single-aliquot-regenerative optically stimulated
144 luminescence (OSL) at 103 ± 8 ka (Murray et al., 2007) and was used in the past as a reference sample to test the accuracy of
145 IR-RF dating (Buylaert et al., 2012). All samples were prepared following conventional treatments (e.g., Preusser et al., 2008),
146 including wet-sieving to isolate the desired grain size fraction, chemical treatment with HCl at 10% to remove carbonates and
147 H_2O_2 at 30% for a few hours to a few days to remove organic matter, and density separation at 2.58 g cm^{-3} using a heavy liquid
148 solution to enrich K-feldspar grains. Sample H22550 was then etched with diluted HF (10%, 40 min). All grains were exposed
149 under a solar simulator SOL Hönle 2 for a few days to reset their signal.

150 Unmeasured grains of sample X7343 were placed on a stub mount on a piece of carbon tape, then imaged with a
151 scanning electron microscope (SEM) equipped for energy-dispersive X-ray spectroscopy (EDS) at Archéosciences Bordeaux
152 (FR) (JEOL JSM-6460LV; detector: Oxford Instruments X-Max (51-XXM0002); software: Oxford Instruments INCA version
153 4.11). The SEM was operated at 20 kV and 55 μ A beam current. Sample X7343 is referred to as BDX22338 in the
154 Archéosciences Bordeaux system.

155 IR-RF measurements were recorded with a *lexsyg research* luminescence reader fitted with an annular $^{90}\text{Sr}/^{90}\text{Y}$ beta
156 source (Richter et al., 2013) using a bandpass filter centred at 850 nm (FWHM 40 nm) mounted in front of a Hamamatsu
157 H7421-50 photo-multiplier tube. Measurements were performed at 70°C , following Frouin et al. (2017). Multi-grains and
158 single-grains were measured on stainless steel cups. High-resolution compositional analysis of the grains was undertaken at
159 the SRX beamline at NSLS-II (Chen-Wiegart et al., 2016). After IR-RF measurements, the grains were removed from the
160 stainless steel cups and fixed on a polymer microscope slide (UVT acrylic; Agar Scientific) with a small piece of carbon tape
161 to avoid misplacement during measurement (supplementary Fig. S1). μ -XRF maps were obtained by scanning across pre-
162 selected regions on the grains with low topographic changes ($90 \times 90 \mu\text{m}$ maps, with a step size of $0.67 \mu\text{m}$ and an integration

163 time of 0.1 s). The incident X-ray beam was focussed by a pair of Kirkpatrick-Baez mirrors. An incident beam energy of 13.5
164 keV was used for the μ -XRF measurements. The excited elements' characteristic fluorescence was obtained from the sum of
165 the four elements of a silicon drift detector. All μ -XRF measurements were normalized to the corresponding incident X-ray
166 flux (I_0) (supplementary Fig. S2). sa.

167 The μ -XANES maps cover 60 x 60 μm in steps of 0.5 μm , thus creating a grid with 120 x 120 data points (i.e., 14
168 400 μ -XRF spectra). To obtain μ -XANES maps of Fe-states in our samples, we varied the incident beam energy according to
169 the absorption edge values obtained from the μ -XANES spectral measurements of Fe standards (Fe foil, pyrite, hematite). The
170 μ -XANES maps were measured three times to obtain μ -XRF emission spectra restricting the Fe species to either (i) the total
171 Fe (at 7.275 keV), (ii) the sum of Fe^{3+} and Fe^{2+} (at 7.134 keV), and (iii) only from Fe^{2+} (at 7.122 keV). The difference between
172 the intensity levels of the latter two measurements can qualitatively give the intensity levels of Fe^{3+} , i.e., $I_{\text{Fe}^{2+} \& \text{Fe}^{3+}} - I_{\text{Fe}^{2+}} =$
173 $I_{\text{Fe}^{3+}}$ where I (a.u.) refers to intensity, thus, the μ -XANES map of Fe^{3+} . We also attempted to record Pb states, however, the Pb
174 standard available at the SRX beamline was fully oxidized, which hindered establishing the correct beam energy for mapping.
175 Therefore, no Pb oxidation state maps were possible. μ -XRF and μ -XANES data were analysed using the open-source software
176 PyXRF v1.0.23 (Li et al., 2017) and ATHENA v0.9.26 (Ravel and Newville, 2005), respectively. Maps and plots were created
177 using **R** (R Core Team, 2022).

178 **4 Results**

179 **4.1 Multi-grain IR-RF signal**

180 The IR-RF signal of a multi-grain aliquot of 8 mm diameter of sample X7343 was first measured. The aliquot contained
181 hundreds of grains. The expected IR-RF signal of K-feldspar grains is a decaying function, e.g., a stretched single-exponential
182 (Erfurt et al., 2003). For sample X7343, however, we observed an unexpected shape of the IR-RF, consisting of a signal
183 decrease until 500 Gy succeeded by an increase, roughly following a saturating exponential shape that keeps increasing beyond
184 ~ 3800 Gy. The regenerative signal for one representative aliquot is shown in figure 1 (top right). We hypothesized that the
185 unexpected signal increase originates from a different source, potentially from a coating around the grains due to the
186 observation of a pinkish/reddish hue on some grains. Clay, Fe-oxide or other grain coatings are a common occurrence and
187 additional preparation steps are sometimes undertaken to remove them prior to luminescence measurements (e.g.,
188 Jayangondaperumal et al., 2012; Lomax et al., 2007; Rasmussen et al., 2023). We attempted to remove this signal
189 contamination, using different chemical treatments such as HF, regal water, and heated regal water, however, without success.
190 Therefore, we decided to investigate the mineral composition of sample X7343, using SEM-EDS on 118 randomly selected
191 grains. Despite using density separation to isolate K-feldspar grains during chemical pre-treatment, we found that this sample
192 was mainly composed of low-K grains (Fig. 1; top left). Indeed, over half of the grains had K-contents less than 2% and less
193 than 5% of the grains had K-contents above 11%. The remainder exhibited K-contents between 2% and 10%. Note that a K-

194 feldspar end member is 14% K (e.g., Gupta, 2015). The low-K grains, which correspond to the majority of grains, also had
195 high Fe-contents of ~10%.

196 We then tested whether it was possible to isolate the desired decreasing IR-RF signal by handpicking grains based on
197 their visual appearance under a microscope. Between 10 to 30 grains were placed onto two aliquots, one for transparent shiny
198 angular grains and one for white-pinkish rounded grains. The regenerated IR-RF signals showed a clear distinction between
199 the two aliquots (Fig. 1), proving it is possible to separate the two observed IR-RF shapes.

200 By manually selecting the grains based on their shape and colour, we made three important observations: i) The
201 decreasing IR-RF signal originates from a small number of grains (less than 5%), presumed to be K-feldspar. ii) The IR-RF
202 signal of these grains decreased beyond 3800 Gy without reaching a plateau, indicating that a dose could be estimated beyond
203 that value. iii) The increasing IR-RF signal originates from a different subset of grains, presumed to be the low-K, Fe-rich
204 minerals identified via SEM-EDS.

205 **4.2 Single-grain IR-RF characterisation**

206 To further investigate this phenomenon, we measured the IR-RF signal of 22 individual grains from five samples of different
207 origins (between one to eight grains per sample, Table 1). Each grain was manually placed on a sample holder (cup) and their
208 signal was recorded over a 3,265 Gy beta irradiation. For each grain, their IR-RF signal shape falls into three categories (Fig.
209 2): Category #1 for grains with a decreasing IR-RF signal, category #2 for grains with an increasing IR-RF signal, and category
210 #3 for grains with a flat signal. Within categories #1 and #2, the saturation level of the individual grains varies (Fig. 2b, d).

211 Among the five samples, one is a tuff, and two are originated from nearby volcanic environments and might, thus, be
212 expected to yield abnormal luminescence behaviour. Common issues with volcanic samples are dim signals, different
213 proportions of emissions, high fading rates and complex grain mineralogy (e.g., Krbetschek et al., 1997; Guérin and Visocekas,
214 2015; Joordens et al., 2015; Sontag-González et al., 2021; O’Gorman et al., 2021). However, we also observed the unwanted
215 increasing IR-RF signal for one of the four grains for sample H22550, which is from a coastal sand deposit. The significance
216 of this find is illustrated in figure 3, where curves representing the signals of individual grains from categories #1 and #2 were
217 added together to simulate a multi-grain aliquot. We used the curves obtained from fitting a single stretched exponential decay
218 function to the normalized data of one grain of sample X7343 (category #2) and one of Gi326 (category #1), since no category
219 #1 grain was measured for sample X7343. When the total signal of the theoretical aliquot was composed of more than 50% of
220 signal from the category #2 grain, we observed the same decay shape as in figure 1 for a multi-grain aliquot of sample X7343.
221 Importantly, a synthetic mixture containing 20% of grains from category #2 still displayed the decaying shape characteristic
222 of category #1 grains. However, the curvature of its dose-response curve was altered, i.e., saturating earlier than the ‘pure’
223 grain. Possible differences in long-term signal stability between the two grain categories could cause differences between the
224 summed curves of natural and regenerated IR-RF signals and thus lead to inaccurate equivalent doses for these mixtures.

225 Further, our results demonstrate that a satisfying IR-RF signal can be measured for all our samples, but only by
226 selecting grains with the appropriate luminescence characteristics (presumably K-feldspar grains). We hypothesize that the

227 low-K grains with high Fe-content are the source of a contaminant IR-RF emission, which if not removed might result in a
228 wrong equivalent dose estimation (i.e., a wrong age estimate).

229 **4.3 Sub-single grain μ -XRF elemental maps**

230 Utilising μ -XRF, we identified up to 15 elements in the grains (see supplementary figure 3 for the total μ -XRF spectra). We
231 then fitted each of the spectra in the 135 by 135 pixel grid (i.e., 18 225 spectra) for each grain to obtain maps describing the
232 XRF intensity of each identified element. These maps only serve as qualitative indicators for the presence of elements and are
233 not corrected for the element-specific emission intensity or the energy-dependent efficiency of the detectors. For two grains,
234 we recorded additional μ -XRF maps to characterize visible inclusions (see table 1).

235 First, we consider only the presence/absence of each element with the IR-RF signals previously obtained. Among the
236 grains displaying a decreasing IR-RF signal (category #1), all contain K, Pb, Fe and Ba, among other elements (Fig. 4, middle).
237 Most of the grains from categories #2 and #3 also contain K, Pb and Fe, but few contain Ba. Further, most grains from
238 categories #2 and #3 contain Ca, Ti and Mn, which are rare in the grains from category #1. The μ -XRF intensities also allow
239 for a qualitative comparison of elemental composition. As shown in the boxplots in figure 4 (right-hand side), category #1
240 grains differ from those in category #2 and #3 primarily by a higher μ -XRF signal contribution from K and a lower contribution
241 from Fe.

242 If we compare the relative intensities of Pb, Fe and K, we can identify a pattern in the composition of grains from
243 each category (Fig. 5). Grains from category #1 tend to have high proportions of K and Pb and medium-to-low proportions of
244 Fe. All grains from categories #2 and #3 have medium-to-high proportions of Fe, and most have low levels of K and Pb. No
245 grains from categories #2 or #3 have high levels of both K and Pb. The element that distinguishes grains from categories #2
246 and #3 is Ca, which is only present in category #2 grains to a high proportion (see also supplementary figure 3).

247 Interestingly, the grains from category #3 cluster relatively close to those from category #2, suggesting that the
248 elemental composition alone is not responsible for the lack of IR-RF signal but that the spatial configuration of the elements
249 is another important factor to consider. This is exemplified in figure 6, which contains elemental maps of K, Pb and Fe for one
250 grain of each category. The category #1 grain (top row) tends to have K and Pb co-localized (overlap shown in green), whereas
251 the category #3 grain (bottom row) also contains both elements, but they appear in separate locations (shown in cyan and
252 yellow). In this grain, K appears co-localized with Fe (overlap shown in dark blue). Our preliminary observations require
253 broader confirmation, but they are in line with the current hypotheses of emission origins for the K-feldspar IR-RF signal
254 decreasing with dose (due to Pb) and a contaminating red RF signal increasing with dose (due to Fe). Furthermore, the lack of
255 an IR-RF signal in category #3 grains appears to stem from low levels of K co-existing with high proportions of Fe, supporting
256 the observations made by Kumar et al. (2020).

257 **4.4 Mapping oxidation states with μ -XANES**

258 By analysing absorption of X-rays near the absorption edge, μ -XANES spectra can provide information on the presence of
259 potential oxidation states of an element, as shown in figure 7 through measurements of different standards of Fe; an increase
260 in the oxidation state is generally accompanied by a shift in the absorption edge to higher energy (Fig. 7, inset).

261 We targeted the Fe-rich region of a category #1 grain (X7368) for mapping (location shown by the blue square in
262 figure 6). Figure 8 shows the μ -XANES maps of three oxidation states (Fe^{3+} and Fe^{2+} combined, Fe^{2+} and Fe^{3+}), all normalized
263 to the maximum intensity level of the total Fe map. These maps suggest that Fe exists in the top section of this feldspar grain
264 in its Fe^{3+} and Fe^{2+} states. Note how Fe^{2+} is mainly clustered in one region, possibly within a mineral inclusion with a rim of
265 Fe^{3+} .

266 **5 Conclusions and future work**

267 We demonstrated that individual K-feldspar grains of the same five samples display different IR-RF behaviour, illustrated by
268 different signal decay shapes (i.e., increasing or decreasing with dose and different saturation levels). These behaviours are
269 cumulative (see figure 1), and therefore, the IR-RF signal of a multi-grain aliquot can lead to inaccurate equivalent doses.
270 Despite the use of chemical preparations to remove contaminants, manually picking individual grains was necessary to isolate
271 K-feldspar grains, which is unrealistic for routine dating applications in a low-light laboratory. A more realistic way to remove
272 such contamination is by selecting K-feldspar grain populations by isolating the emission signal of individual grains with an
273 imaging system. Here, we wanted to gain a further understanding of the production and origin of the emission signal, which
274 ultimately will help us design a more appropriate imaging system for IR-RF dating. For sample X7343, we show through
275 SEM-EDS analyses that the different emissions can be linked to different grain mineralogy. Since K-feldspar grains are known
276 to be heterogeneous on a subgrain level, we propose synchrotron-based X-ray spectroscopy to characterize the grains on a
277 submicron scale and investigate the origin of the IR-RF and other linked emissions. Information on the oxidation states of,
278 e.g., Fe possibly allows for the characterisation of the reactions behind the electronic changes leading to radiofluorescence.

279 In the preliminary work presented here, we successfully applied μ -XRF and μ -XANES at the SRX beamline (NSLS-
280 II) to obtain element and oxidation state maps of regions of interest within individual K-feldspar coarse grains previously used
281 for IR-RF measurements. We were able to correlate the desired IR-RF signal shape (category #1) with compositions of high
282 proportions of K, Pb, and Ba and low proportions of Fe. High proportions of Fe in the μ -XRF spectra were found in grains of
283 categories #2 and #3, but the possible role of Fe as a contaminant remains unclear. During our next beam time, we will polish
284 the grains down to a uniform surface prior to μ -XRF and μ -XANES measurements to limit surface effects. Such a setup will
285 also allow us to test the hypothesis that the contaminating IR-RF signal is coming from an element present at the surface of
286 the grain (e.g., iron coating possibly due to weathering), but not within the grain.

287 The relation between the chemical composition, crystal structure, and the shape of the IR-RF signal in individual K-
288 feldspar grains is still poorly understood, and efforts should be made to identify and quantify at high resolution the element

289 responsible for producing the IR-RF signal with the highest dynamic range (i.e., saturation at high dose). Our future work will
290 include implementing a second detector to simultaneously measure μ -XRF/ μ -XANES and the IR-RF signal induced by the X-
291 rays. Though not widely used, X-rays are a suitable alternative to radioactive sources for luminescence dosimetry including
292 RF. The dual detection will allow us to isolate emissions from different mineral inclusions and directly correlate them to the
293 elemental composition, thereby assessing the extent of overlap of the desired IR-RF emission centred at 880 nm and
294 contaminating ones such as the possible unstable red emission associated with Fe^{3+} .

295 **Data availability**

296 The SEM dataset and the original data used to produce μ -XRF maps are available online (Sontag-González et al., 2023).

297 **Author contribution**

298 MF, JT, RK and JLS designed the experiments and prepared the samples. RK, JLS, and MF carried out the IR-RF
299 measurements. SK organized and analysed the SEM EDS measurements. MF and JT carried out the μ -XRF and μ -XANES
300 measurements. RK, MSG and MF analysed the results. MSG and RK prepared the manuscript with contributions from all
301 authors. MF, JT and JLS obtained funding.

302 **Competing interests**

303 The authors declare that they have no conflict of interest.

304 **Acknowledgements**

305 We are grateful to Svenja Riedesel and an anonymous referee for very constructive comments on an earlier version of this
306 manuscript. We thank Yannick Lefrais for operating the EDS at Archéosciences Bordeaux (former IRAMAT-CRP2A) in 2018.
307 This research used the SRX beamline of the National Synchrotron Light Source II, a U.S. Department of Energy (DOE) Office
308 of Science User Facility operated for the DOE Office of Science by Brookhaven National Laboratory under Contract No. DE-
309 SC0012704.

310 **Financial support**

311 This work was supported by the Natural Environment Research Council (grant number NE/T001313/1); and a Stony Brook
312 University-Brookhaven National Laboratory Seed Grant (#94508). The SEM analysis at Archéosciences Bordeaux was
313 supported by the Agence Nationale de la Recherche (grant no. ANR-10-LABX-52).

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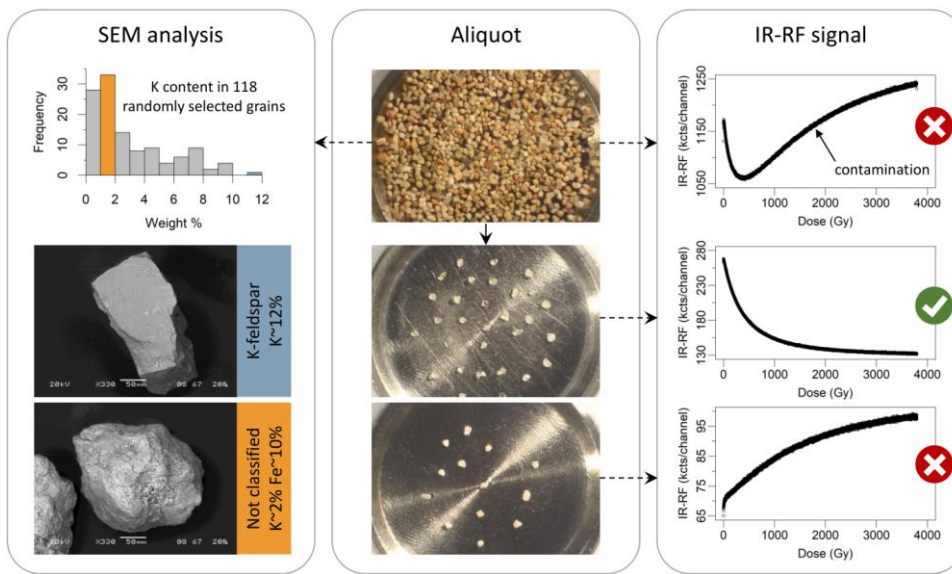
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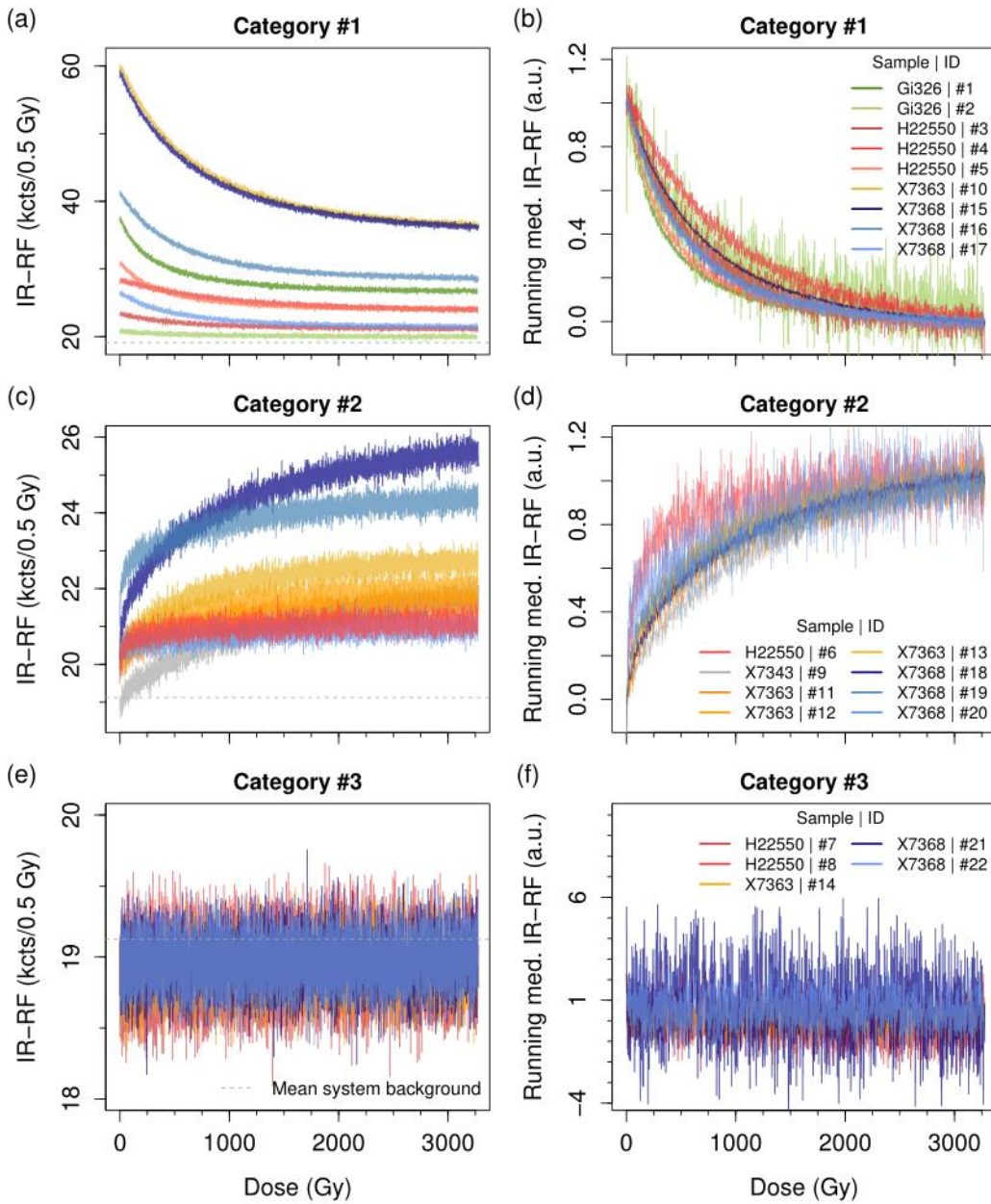
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Figure 1: Illustration showing how contamination of the IR-RF signal can be removed by selecting only K-feldspar grains from sample X7343. The regenerative IR-RF curves were obtained from aliquots containing hundreds of unsorted grains (top) or 10–30 grains manually sorted into transparent shiny angular grains (middle) or white-pinkish rounded grains (bottom). The histogram shows the K-content determined by SEM-EDS for 118 grains (not measured for IR-RF). Representative examples of grains classified as K-rich and contaminating Fe-rich grains are shown.



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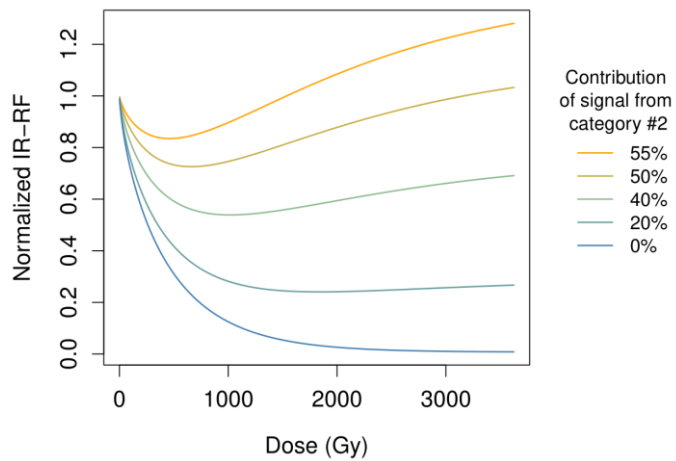
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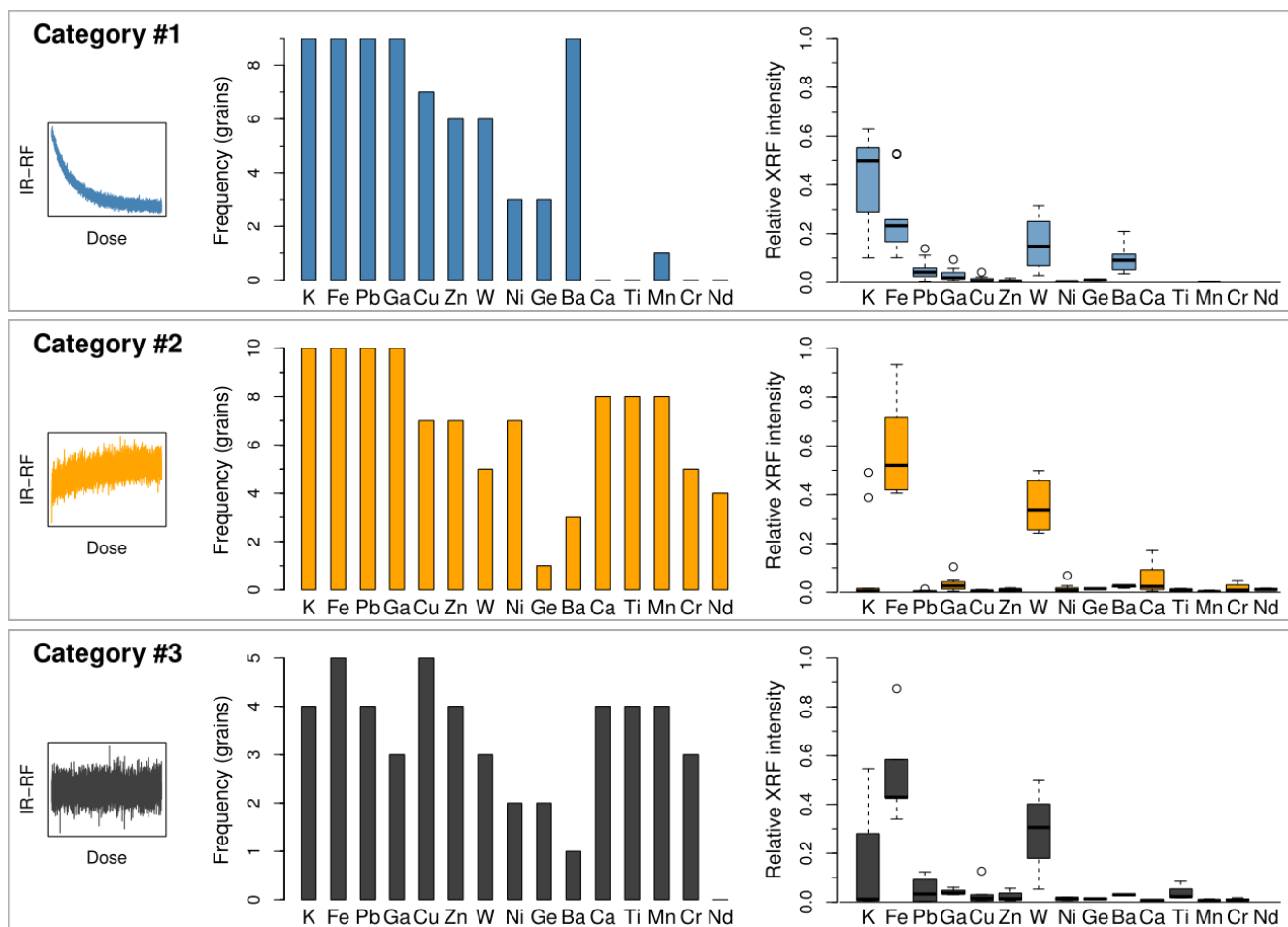
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Figure 2: IR-RF dose-response curves of individual grains obtained after bleaching. Categories #1–#3 refer to grains with decreasing, increasing or no detectable signal, respectively (one category per row). The curves are shown (a, c, e) unnormalized and without background correction and (b, d, f) with intensities normalized to the signal maxima (defined as the median value of (b, f) the initial and (d) the final 20 channels) after subtracting as background the minimum signal of each grain (defined as the median value of (b, f) the final and (d) the initial 20 channels). For better visualisation, the normalized plots show the running median IR-RF with a window of 7 values. The system background was determined as the mean value obtained from measuring 5 empty cups under the same conditions as the grains.



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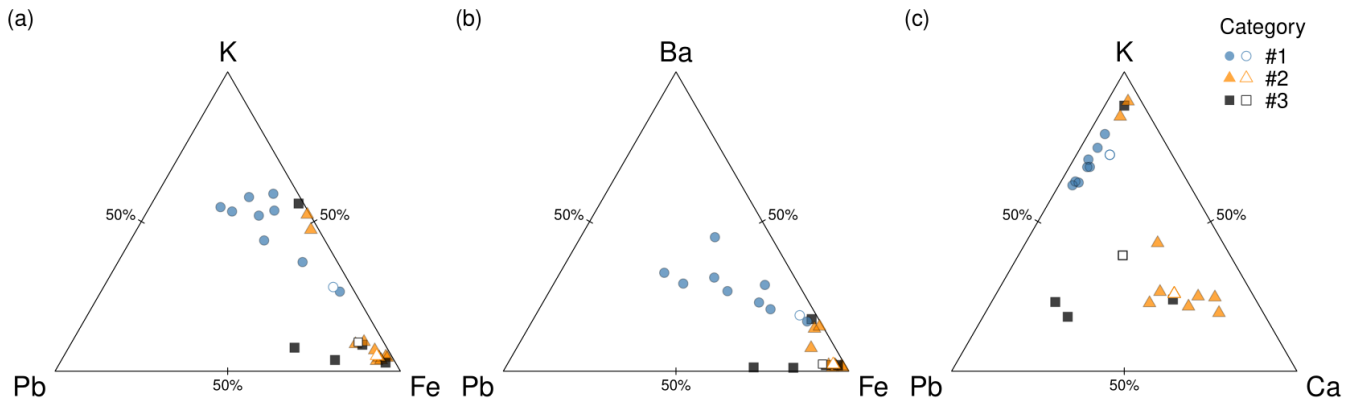
469 **Figure 3:** Simulated dose-response curves of theoretical aliquots varying the proportion of grains from categories #1 (desired decreasing
 470 signal) and #2 (increasing signal). The curves are the sum of two stretched exponentials using parameters obtained from fits of grains from
 471 samples Gi326 (category #1) and X3743 (category #2). The higher the signal contribution from category #2 grains, the more aberrant the
 472 sum curve becomes.



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475 **Figure 4:** Bar charts of elements identified in μ -XRF spectra and boxplots of the relative μ -XRF intensities for grains in three categories, as
 476 as exemplified in the insets: decreasing IR-RF signal (category #1), increasing IR-RF signal (category #2) or flat IR-RF signal indistinguishable
 477 from the background (category #3) during beta irradiation.

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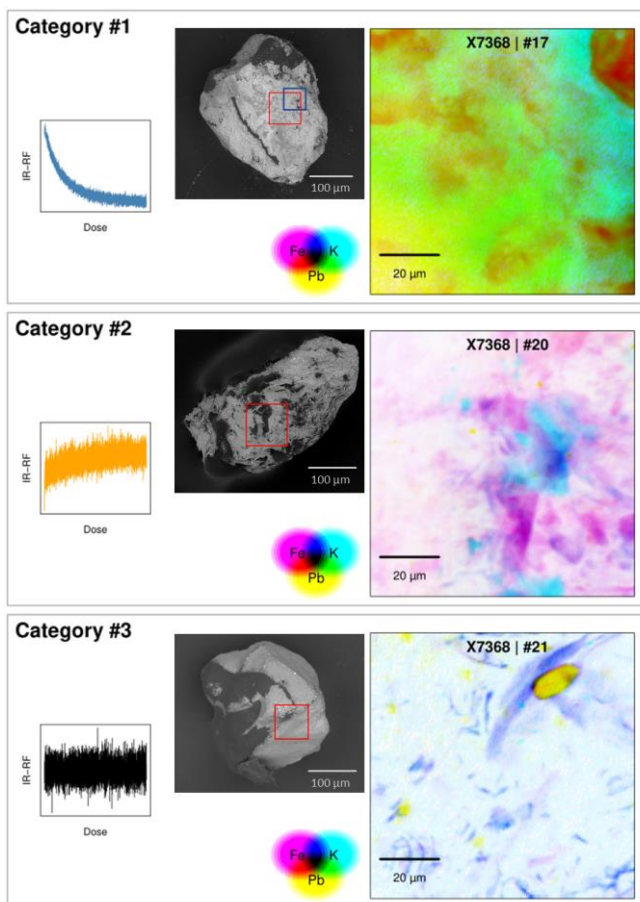
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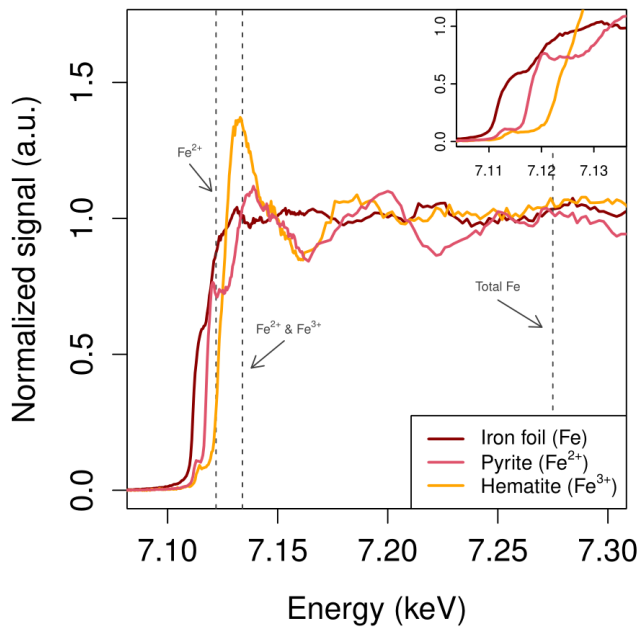
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Figure 5: Ternary diagrams of relative μ -XRF intensities attributed to (a) K, Fe and Pb, (b) Ba, Fe and Pb, and (c) K, Ca and Pb for grains of the three categories. Note that the contributions are not calibrated to mass or stoichiometry. The relative K contribution is, thus, not directly comparable to the K-feldspar K-content. The three grains shown in figure 6 are marked as open symbols in each ternary diagram.

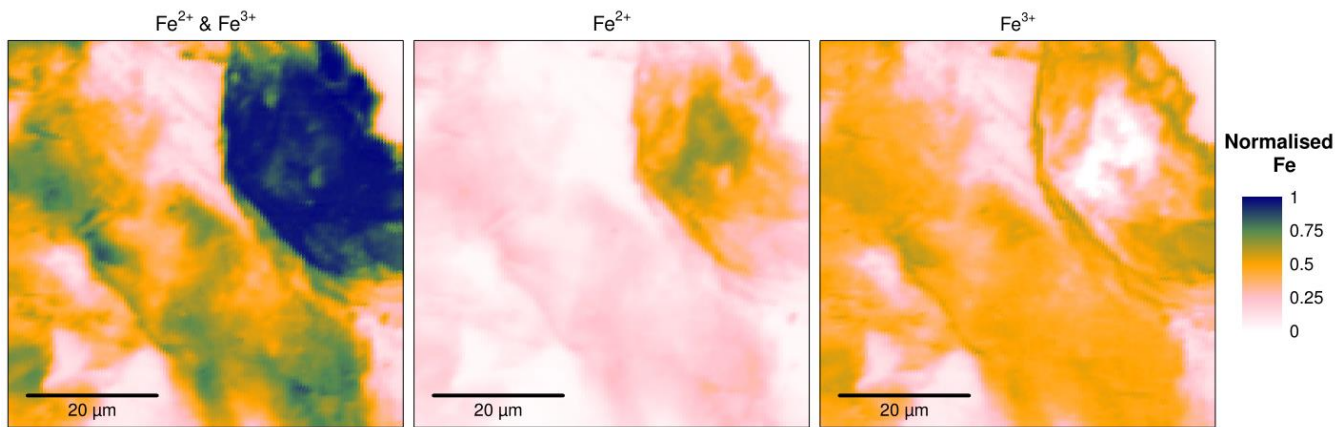


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 484 **Figure 6:** Illustration showing three IR-RF curves obtained from three grains of sample X7368, classified as follows: decreasing IR-RF
 485 signal (category #1), increasing IR-RF signal (category #2) or flat IR-RF signal indistinguishable from the background (category #3) during
 486 beta irradiation. μ -XRF spectra were measured from the area bordered by red squares on the SEM images of the grains. The maps show the
 487 presence of K, Fe, and Pb on the same grains as the IR-RF curves. The elemental compositions are shown overlaid, with the colour scales
 488 normalized to the maximum contribution of each element for each grain. The area bordered by a blue square in the category #1 grain
 489 corresponds to the map shown in figure 8.



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Figure 7: μ -XANES spectra of Fe standards. The dashed vertical lines indicate the incident beam energies necessary to isolate emissions from specific oxidation states. The inset shows a magnification of the energy region relevant to determine the incident beam energies.



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Figure 8: Maps of Fe oxidation states for a grain of sample X7368 (category #1; ID #17). Intensities are normalized to the maximum intensity of total Fe.

498 **Table1:** Overview of measured grains. Categories #1–#3 refer to grains with decreasing, increasing or no detectable signal, respectively.
499 For two grains, two regions each were mapped, so we measured a total of 24 μ -XRF maps.

Sample	Grain size (μm)	Number of measured grains		
		Category #1	Category #2	Category #3
Gi326	90–200	2	0	0
H22550	180–250	3	1*	2
X7343	180–255	0	1*	0
X7363	180–255	1	3	1
X7368	180–255	3	3	2
Total		9	8	5

500 *For these grains, two regions were mapped by μ -XRF: the grain ‘matrix’ and an inclusion.