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Short communication: Synchrotron-based elemental mapping of

single grains to investigate variable infrared-radiofluorescence

emissions for luminescence dating

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Abstract. During ionising irradiation, potassium (K)-rich feldspar grains emit infrared (IR) light, which is used for infraredradiofluorescence (IR-RF) dating. The late-saturating IR-RF emission centred at ~880 nm represents a promising tool to date

- 20 Quaternary sediments. However, in the present work, we report the presence of individual grains of K-feldspar displaying an aberrant IR-RF signal shape, whose combined intensity contaminates the sum signal of an aliquot composed of dozens of grains. Our experiments were carried out at the National Synchrotron Light Source (NSLS-II) on coarse (> 90 µm) K-feldspar grains of five samples of different ages, nature and origin in order to characterise the composition of grains yielding the desired or contaminated IR-RF emission. Using micro-X-ray-fluorescence (µXRF), we successfully acquired element distribution
- 25 maps of fifteen elements (<1 µm resolution) of the surface of grains previously used for luminescence dating. In keeping with current theories of IR-RF signal production, we observed a correlation between the relative proportions of Pb and Fe and the shape of the luminescence signal: most grains with the desired IR-RF signal shape had high Pb and low Fe contents. Interestingly, these grains were also defined by high Ba and low Ca contents. Additionally, this study represents a proof-of-concept for mapping the oxidation states of Fe-ions using micro-X-ray absorption near-edge structure spectroscopy (µXANES)</p>
- 30 on individual grains. The high spatial resolution enabled by synchrotron spectroscopy makes it a powerful tool for future experiments to elucidate long-standing issues concerning the nature and type of defect(s) associated with the main dosimetric trap in feldspar.





1 Introduction

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Geochronologic data provide essential information for understanding the rates of Earth's surface processes, environmental changes, and the evolution of life. Advances in dating techniques have fundamentally changed our capacity to piece together our evolutionary past over millions of years, with luminescence dating proving a powerful tool in this field as it applies to various types of sediments and contexts. The technique determines an age estimate for when mineral grains were last exposed to daylight or heat. Luminescence dating methods rely on the capacity of certain minerals to record the amount of radiation to which they have been exposed during burial and release energy when exposed to sunlight or high temperature (Aitken, 1985, 40 1998; Bateman, 2019). In the laboratory, the total amount of energy stored in the mineral is measured as a dose (Gy). The energy absorption rate (dose rate, Gy a⁻¹) is derived from knowledge of the natural radioactivity surrounding the sampled sediment. The quotient of these two values (dose/dose rate) gives the burial time.

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Generally, the preferred mineral is quartz because of its high abundance and resistance to weathering. However, the early saturation of the optically stimulated luminescence (OSL, Huntley et al., 1985) signal within quartz at ~200 Gy (Wintle and Adamiec, 2017) generally limits its application to the last 200,000 years (considering a low dose rate of 1 Gy ka⁻¹). By contrast, potassium (K-) rich feldspar minerals typically display significantly higher dose saturation levels around 600 Gy (see summary in Sec. 8.1 in Murari et al., 2021a) using infrared stimulated luminescence (IRSL, Hütt et al., 1988), allowing the dating of older deposits up to 600,000 years (considering a low dose rate of 1 Gy ka⁻¹).

Over the past decades, different methods have been proposed to extend this upper age limit with varying degrees of 50 success. The infrared-radiofluorescence (IR-RF) signal of K-feldspar is a promising candidate for such an extension. The IR-RF emission at 880 nm (Kumar et al., 2018; Riedesel et al., 2021; Sontag-González et al., 2022) arises from prompt radiative recombination of charge within crystalline materials during continuous exposure to ionizing radiation. The IR-RF signal decreases in intensity with the dose accumulation as the electron traps fill until saturation (Trautmann et al., 1999a). This saturation level constrains the time range over which IR-RF dating is applicable. The IR-RF dose is determined in three 55 measurement steps: (i) an additive irradiation is given while the natural IR-RF signal is recorded, (ii) the sample is optically bleached to empty the traps and after a pause, (iii) a second irradiation is performed during which the regenerated IR-RF signal is measured. Finally, the natural dose absorbed during burial is calculated from the horizontal distance over which the natural signal needs to be shifted to match the regenerative signal curve (see Murari et al., 2021a for a review).

Murari et al. (2018) demonstrated that accurate dose recovery of a known dose of 3600 Gy is possible (a dose recovery 60 test is a laboratory performance check of the measurement protocol, and successful dose recovery is a prerequisite for any protocol). If we assume typical environmental dose rates of between 3 Gy ka⁻¹ and 1 Gy ka⁻¹, then IR-RF dating could produce age estimates ranging from 1.2 Ma to 3.6 Ma, which is around four times greater than the upper dating limit of conventional luminescence dating methods. However, more recent studies (Murari et al., 2021b; Kreutzer et al., 2022) indicated a dose saturation cap at around 1500 Gy, reducing the previously assumed temporal limit of IR-RF dating. Hence, the uncertainty 65 surrounding its upper age limit remains and further studies on known-age samples are required to assess whether the





sample/grain geochemistry influences the age limit. There is undoubtedly a gap in our current understanding of the luminescence production processes in K-rich feldspar, and a revised conceptual model might be needed.

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To assess whether the grain geochemistry influences the IR-RF signal and perhaps the age limit of IR-RF, we examined individual K-feldspar grains at the submicron resolution X-ray spectroscopy (SRX) beamline at the National Synchrotron Light Source II (NSLS-II) at Brookhaven National Laboratory. Measurements at such a high-resolution will lead to a better understanding of the luminescence kinetics in feldspars. Here we report on the feasibility and practicality of using μ -X-ray fluorescence (μ -XRF) and μ -X-ray absorption near-edge fine structure (μ -XANES) techniques in investigating the luminescence signal origin and kinetic in K-feldspar.

2 Method rationale

75 Trautmann et al. (2000) were the first to analyse the IR-RF signal of individual K-feldspar grains. Using spectral measurements on twenty-one to forty-two grains from three samples, they observed up to four emissions (IR, red, yellow, blue) with variable intensities (a fourth sample appears in their figure 3 but is not mentioned in the main text). An IR-RF dose-response curve was only reported for one grain, which had a similar shape, albeit a later onset of saturation, when compared to the response from the multi-grain aliquot of the same sample. More recently, Mittelstraß and Kreutzer (2021) analysed sixty grains from two 80 samples, of which 55% and 80% emitted a detectable signal. In that study, between one and three grains (~9% of signalemitting grains for both samples) were rejected due to a bad match between the natural and regenerative curves. However, all grains that emitted a detectable signal displayed the expected decay shape. Likewise, our laboratory observations indicate that the signal varies in sensitivity across feldspar minerals and can be contaminated for various reasons, leading to spectral interference or quenching, ultimately influencing the saturation level and/or the shape of the IR-RF signal (Frouin et al., 2017, 85 2019; Kumar et al., 2020).

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Previous publications placed the IR emission at 1.43 eV (865 nm) based on Trautmann et al. (1999a, b) and Erfurt and Krbetschek (2003), but more recent work including corrections for the spectrometer efficiency places the IR emission closer to 880 nm (Kumar et al., 2018; Riedesel et al., 2021; Sontag-González et al., 2022). A second IR emission centred at 955 nm (1.30 eV) at lower intensity has also been identified (Kumar et al., 2018). With dose exposure, the 955 nm emission increases and overlaps with the 880 nm peak.

Previous spectral analyses of K-rich feldspar indicated that the IR-RF emission occurs as a result of the change in the

oxidation state of the participating defect via the transition: $Pb^{2+} \rightarrow (Pb^+)^* \rightarrow Pb^+$ (Nagli and Dyachenko, 1986; Erfurt, 2003). A similar transition has been suggested for Amazonite (see Ostrooumov, 2016), but the direct connection between the Pbcentre and IR-RF is not yet evidenced. Similarly, the presence of Fe in feldspar is also known to lead to a red RF emission (e.g., Telfer and Walker, 1978; Brooks et al., 2002; Visocekas et al., 2014), with the maximum peak wavelength varying 95 between 700 nm and 770 nm depending on feldspar composition (Krbetschek et al., 2002). Despite the occurrence of the red RF emission in Fe³⁺ state, its precise origin remains a subject of debate, with conflicting opinions suggesting Fe²⁺ (Fe²⁺ + h \rightarrow





 Fe^{3+} ; here h stands for hole) or Fe^{4+} ($Fe^{4+} + e^{-} \rightarrow Fe^{3+}$; here e⁻ stands for electron) as the potential sources (Kumar et al., 2020). Furthermore, spectral analyses showed that, with dose exposure, the red RF emission (~710 nm emission in K-feldspar) 100 increases, while the 880 nm emission decreases (Krbetschek et al., 2000; Erfurt and Krbetschek, 2003; Kumar et al., 2018; Frouin et al., 2019). The thermal stability of the ~710 nm emission has been, however, questioned (Krbetschek et al., 2000). Such a reduced thermal stability might be an issue for IR-RF dating, as it has been suggested that the tail of the ~710 nm emission overlaps with the 880 nm emission, thus potentially playing a role in the shape of the measured IR-RF. Such a contribution can be reduced to less than 5% of the IR-RF signal by using a bandpass filter centred at 850 nm (FWHM 40 nm)(see Sontag-González and Fuchs, 2022). In summary, although previous studies have identified factors that may influence the IR-RF signal in several ways, e.g., whether the IR-RF signal originates from Pb, and is affected by the presence of Fe²⁺ or Fe⁴⁺, a conclusive confirmation or comprehensive linkage between these factors is yet to be established.

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Identification of the defect type linked to the IR-RF signal and its concentration will enable us to better characterise the light emission (signal sensitivity) in different types of feldspar, while identification of the origin of possible contamination in the IR-RF signal will help us to gain a better understanding of the apparent early saturation or quenching of the IR-RF signal. The µ-XRF and µ-XANES techniques are best suited for this purpose by producing high-resolution maps of elements and their oxidation states. The use of synchrotron μ XRF allows us to improve the spatial resolution compared with previous uses of μ XRF (e.g., Buylaert et al., 2018) by reducing the beam spot size from ~25 μ m to 1 or 0.5 μ m. Though both the grain geochemistry and crystallography should be investigated to characterise the defect type and its environment, in the present study, we decided to focus only on geochemistry.

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First, we recorded IR-RF curves from individual grains in our luminescence dating laboratory at the Research Laboratory for Archaeology and the History of Art (RLAHA) at the University of Oxford (UK). Then, during our beamtime (96 h), and as a proof of concept, we optimised the measurement conditions and obtained compositional maps of the individual K-feldspar grains. We paid particular attention to K, Ca, Fe and Pb. After analysing the μ -XRF maps, μ -XANES measurements were done at selective spots where Fe and Pb occur in greater concentrations. The atomic number of sodium (Na; another end member of the feldspar ternary system) is too low to be measured at the current SRX beamline.

A total of five samples were selected to represent a diversity of i) geological context, ii) geochemistry, iii) shape of the IR-RF

3 Material and instrumentation

signal, and iv) age. Sample Gi326 from a Triassic sandstone from Bayreuth, Germany is composed of 89% of K-rich feldspar 125 (Sontag-González and Fuchs, 2022) and has previously been used as a reference sample in a laboratory comparison of IR-RF dating (Murari et al., 2021b). X7343 was collected from a Pliocene sediment at the Nyayanga site in the Homa peninsula, Kenya (Plummer et al., 2023). X7363 was taken from the Gele Tuff in the Turkana Basin, Kenya, and dated by Argon-Argon at 1.32 ± 0.2 Ma (Phillips et al., 2023). X7368 is a sediment sample collected above the Silbo Tuff (0.751 \pm 0.022 Ma, McDougall and Brown, 2006) and below the Kale Tuff (undated) in the Turkana Basin, Kenya. Sample H22550 is a coastal





130 marine sample from Sula, Russia, dated by quartz single-aliquot-regenerative OSL at 103 ± 8 ka (Murray et al., 2007) and used as a reference sample to test the accuracy of IR-RF (Buylaert et al., 2012). All samples were prepared following conventional treatments (e.g., Preusser et al., 2008), including wet-sieving to isolate the desired grain size fraction, chemical treatment with HCl at 10% to remove carbonates and H₂O₂ at 30% for a few hours to a few days to remove organic matter, and density separation at 2.58 g cm⁻³ using a heavy liquid solution to enrich K-rich feldspar grains. Sample H22550 was then etched with diluted HF (10%, 40 min). All the grains were exposed under a solar simulator SOL Honle 2 for a few days to

reset their signal.

Grains of sample X7343 were placed on a stub mount on a carbon tape, then imaged with a scanning electron microscope (SEM) equipped for energy-dispersive X-ray spectroscopy (EDS) at Archéosciences Bordeaux, France (JEOL JSM-6460LV; detector: Oxford Instruments X-Max (51-XMX0002); software: Oxford Instruments INCA version 4.11). The

140 SEM was operated at 20 kV voltage and 55 μA beam current. Sample X7343 is referred to as BDX22338 in the Archéosciences Bordeaux system.

IR-RF measurements were recorded with a Lexsyg research luminescence reader fitted with an annular ⁹⁰Sr/⁹⁰Y beta source (Richter et al., 2013) using a bandpass filter centred at 850 nm (FWHM 40 nm) mounted in front of a Hamamatsu H7421-50 photo-multiplier tube. Measurements were performed at 70°C, following Frouin et al. (2017). Multi-grain and

- 145 single-grains were measured on stainless steel cups. High-resolution compositional analysis of the grains was undertaken at the 5-ID SRX beamline at the National Synchrotron Light Source II (NSLS-II) at Brookhaven National Laboratory (Chen-Wiegart et al., 2016). Grains were fixed on a polymer microscope slide with a small piece of carbon tape to avoid misplacement during measurement (supplementary figure 1). XRF maps were obtained by scanning across pre-selected regions on the grains (90 x 90 µm maps, integration time: 0.1 s). A resolution of 0.67 µm was achieved by focusing the beam with a pair of
- 150 Kirkpatrick-Baez mirrors. An incident beam energy of 13.5 keV was used for the XRF measurements. The excited elements' characteristic fluorescence was detected through the sum of 4 silicon drift detectors. All XRF measurements were normalised to the corresponding incident X-ray flux (I₀) (supplementary figure 2). The XANES maps had a resolution of 0.5 μm (60 x 60 μm). To do μXANES mapping of Fe-states in our samples, we varied the incident beam energy according to the absorption edge values obtained from the μXANES measurements of Fe standards (Fe foil, pyrite, hematite). The μXANES maps were
- 155 measured three times to obtain μ XRF emission spectra restricting the Fe species to either (i) the total Fe (at 7.275 keV), (ii) the sum of Fe³⁺ and Fe²⁺ (at 7.134 keV) and (iii) only from Fe²⁺ (at 7.122 keV). The difference between the intensity levels of the latter two measurements can qualitatively give the intensity levels of Fe³⁺, i.e., $I_{Fe2+\& Fe3+} - I_{Fe2+} = Fe3+$ where *I* refers to intensity, thus, the μ XANES map of Fe³⁺. We also attempted to record Pb states, however, the Pb standard available at 5-ID was fully oxidised, which hindered establishing the correct beam energy for mapping. Therefore, no Pb oxidation state maps
- 160 were possible. XRF and XANES data were analysed using the open-source software PyXRF v.1.0.23 (Li et al., 2017) and ATHENA v.0.9.26 (Ravel and Newville, 2005), respectively.





4 Results

4.1 Multi-grain IR-RF signal

- The IR-RF signal of a multi-grain aliquot of 8 mm diameter of sample X7343 was first measured. The aliquot contained 165 hundreds of grains. The expected IR-RF signal of K-rich feldspar grains is a decaying function, e.g., a stretched singleexponential (Erfurt et al., 2003). For sample X7343, however, we observed an unexpected shape of the IR-RF measurements, consisting of a signal decrease until 500 Gy succeeded by an increase, roughly following a saturating exponential shape that keeps increasing beyond ~3800 Gy. The regenerative signal for one representative aliquot is shown in figure 1 (top right). To remove this signal contamination, potentially coming from coating around the grains, we used different chemical treatments
- 170 such as HF, regal water, and heated regal, however, without success. Therefore, we decided to investigate the mineral composition of sample X7343, using SEM-EDS on a hundred randomly selected grains. Despite using density separation to isolate K-rich feldspar grains during chemical pre-treatment, we found that this sample was mainly composed of low-K grains (Fig. 1; top left). Indeed, over half of the grains had K-contents less than 2% and less than 5% of the grains had K-contents above 11%. The remainder had K-contents between 2 and 10%. Note that a K-feldspar end member is 14% K (e.g., Gupta, 175 2015). The low-K grains, which correspond to the majority of grains, also had high Fe-contents of $\sim 10\%$.

We then tested whether it was possible to isolate the desired decreasing IR-RF signal by handpicking grains based on their visual appearance under a microscope. Between ten to thirty grains were placed onto two aliquots, one for transparent shiny angular grains and one for white-pinkish rounded grains. The regenerated IR-RF signals showed a clear distinction between the two aliquots (Fig. 1), proving it is possible to separate the two observed IR-RF shapes.

180 By manually selecting the grains based on their shape and colour, we made three important observations: i) The decreasing IR-RF signal originates from a small number of grains (less than 5%), presumed to be K-rich feldspar. ii) The IR-RF signal of these grains decreases beyond 3800 Gy without reaching a plateau, indicating that a dose could be estimated beyond that value. iii) The increasing IR-RF signal originates from a different subset of grains, presumed to be the low-K, Ferich minerals identified via SEM-EDS.

185 4.2 Single-grain IR-RF characterisation

To further investigate this phenomenon, we measured the IR-RF signal of twenty-two individual grains coming from five samples from different origins (between one to eight grains per sample, Table 1). Each grain was manually placed on a sample holder (cup) and their signal was recorded over a 3265 Gy beta irradiation. For each grain, their IR-RF signal shape falls into three categories (Fig. 2): Category #1 for grains with a decreasing IR-RF signal, category #2 for grains with an increasing IR-

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RF signal, and category #3 for grains with a flat signal. Within categories #1 and #2, the saturation level of the individual grains varies (Fig. 2b, d).

Among the five samples, one is a tuff, and two are originated from nearby volcanic environments and might, thus, be expected to yield abnormal behaviour. However, we also observed the unwanted decreasing IR-RF signal for one of the four





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grains for sample H22550, which is from a coastal sand deposit. The significance of this find is illustrated in figure 3, where curves representing the signals of individual grains from categories #1 and #2 were added together to simulate a multi-grain aliquot. We used the curves obtained from fitting a single stretched exponential decay function to the normalised data of one grain of sample X7343 (category #2) and one of Gi326 (category #1), since no category #1 grain was measured for sample X7343. When the total signal of the theoretical aliquot was composed of more than 50% of signal from the category #2 grain, we observed the same decay shape as in figure 1 for a multi-grain aliquot sample X7343. Importantly, a synthetic mixture 200 containing 20% of grains from category #2 still displayed the decaying shape characteristic of category #1 grains. However, the curvature of its dose-response curve was altered, i.e., saturating earlier than the 'pure' grain. Possible differences in longterm signal stability between the two categories could cause differences between the total curves of natural and regenerated IR-RF signals and thus lead to inaccurate equivalent doses for these mixtures.

Further, our results demonstrate that a satisfying IR-RF signal can be measured for all our samples, but only by 205 selecting K-rich feldspar grains, while other (presumed) low-K grains can contaminate the IR-RF emission and might result in a wrong equivalent dose estimation (i.e., a wrong age estimate).

4.3 Subgrain µXRF elemental maps

We identified fifteen elements in the grains (see supplementary figure 3 for the total spectra). We then fitted each of the 18 225 spectra for each grain to obtain maps describing the XRF intensity of each identified element. These maps only serve as qualitative indicators for the presence of elements and do not consider the element-specific emission intensity or the energydependent efficiency of the detectors. For two grains, we recorded additional µXRF maps to characterise visible inclusions (see Table 1).

First, we consider only the presence/absence of each element with the IR-RF signals previously obtained. Among the grains displaying a decreasing IR-RF signal (category #1), all contain K, Pb, Fe and Ba, among other elements (Fig. 4). Most 215 of the grains from categories #2 and #3 also contain K, Pb and Fe, but few contain Ba. Further, most grains from categories #2 and #3 contain Ca, Ti and Mn, which are rare in the grains from category #1.

The XRF intensities allow for a qualitative comparison of elemental composition. If we compare the relative intensities of Pb, Fe and K, we can identify a pattern in the composition of grains from each category (Fig. 5). Grains from category #1 tend to have high proportions of K and Pb and medium-to-low proportions of Fe. All grains from categories #2 and #3 have medium-to-high proportions of Fe, and most have low levels of K and Pb. No grains from categories #2 or #3 have high levels of both K and Pb. The element that distinguishes grains from categories #2 and #3 is Ca, which is only present

Interestingly, the grains from category #3 cluster relatively close to those from category #2, suggesting that the elemental composition alone is not responsible for the lack of IR-RF signal but that the spatial configuration of the elements

225 is another important factor to consider. This is exemplified in figure 6, which contains elemental maps of K, Pb and Fe for one grain of each category. The category #1 grain (top row) tends to have K and Pb co-localised (overlap shown in green), whereas

in category #2 grains to a high proportion (see supplementary figure 3).



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the category #3 grain (bottom row) also contains both elements, but they appear in separate locations (shown in cyan and yellow). In fact, in this grain, K appears co-localised with Fe (overlap shown in dark blue). Our preliminary observations must still be confirmed for a larger number of grains, but they are in line with the current hypotheses of emission origins for the K-feldspar IR-RF signal decreasing with dose (Pb) and a contaminating red RF signal increasing with dose (Fe).

4.4 Mapping oxidation states with µXANES

By analysing absorption of X-rays near the absorption edge, μ XANES spectra can provide information on the presence of potential oxidation states of an element, as shown in figure 7 through measurements of different standards of Fe; an increase in the oxidation state is generally accompanied by a shift in the absorption edge to higher energy (Fig. 7, inset).

235 We targeted the Fe-rich region of a category #1 grain (X7368) for mapping. Figure 8 shows the μ XANES maps of three oxidation states (Fe³⁺ and Fe²⁺ combined, Fe²⁺ and Fe³⁺), all normalised to the maximum intensity level of the total Fe map. These maps suggest that Fe exists on the surface of this feldspar grain in its Fe³⁺ and Fe²⁺ states. Note how Fe²⁺ is mainly clustered in one region, possibly within a mineral inclusion with a rim of Fe³⁺.

5 Conclusions and future work

- 240 We demonstrated that individual K-feldspar grains of the same five samples display different IR-RF behaviour, illustrated by different signal decay shapes (i.e., increasing or decreasing with dose and different saturation levels). These behaviours are cumulative (see figure 1), and therefore, the IR-RF signal of a multi-grain aliquot can lead to inaccurate equivalent doses. Despite the use of chemical preparations to remove contaminants, manually picking individual grains was necessary to isolate K-feldspar grains, which is unrealistic for routine dating applications in a low-light laboratory. A more realistic way to remove
- such contamination is by selecting K-rich feldspar grain populations by isolating the emission signal of individual grains with an imaging system. Here, we wanted to gain a further understanding of the production and origin of the emission signal, which ultimately will help us design a more appropriate imaging system for IR-RF dating. For sample X7343, we show through SEM-EDS analyses that the different emissions can be linked to different grain mineralogy. Since K-feldspar grains are known to be heterogeneous on a subgrain level, we propose synchrotron-based X-ray spectroscopy to characterise the grains on a submicron scale and investigate the origin of the IR-RF and other linked emissions. Information on the oxidation states of,
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e.g., Fe and Pb, possibly allow for the characterisation of the reactions behind the electronic changes leading to radiofluorescence.

In the preliminary work presented here, we successfully applied µXRF and µXANES at the SRX beamline (NSLS-II) to obtain mineralogical and oxidation state maps of regions of interest within individual K-feldspar coarse grains previously used for IR-RF measurements. We were able to correlate the desired IR-RF signal shape (category #1) with compositions of high proportions of K, Pb, and Ba and low proportions of Fe. High proportions of Fe in the µXRF spectra were found in grains of categories #2 and #3, but the possible role of Fe as a contaminant remains unclear. During our next beamtime, we will polish



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the grains down to a uniform surface prior to μ XRF and μ XANES measurements to avoid surface effects. Such a setup will also allow us to test the hypothesis that the contaminating IR-RF signal is coming from an element present at the surface of the grain (e.g., iron coating possibly due to weathering), but not within the grain.

The relation between the chemical composition, crystal structure, and the shape of the IR-RF signal in individual Kfeldspar grains is still poorly understood, and efforts should be made to identify and quantify at high resolution the element responsible for producing the IR-RF signal with the highest dynamic range (i.e., saturation at high dose). Our future work will include implementing a second detector to simultaneously measure μ XRF/ μ XANES and the IR-RF signal induced by the Xrays. Though not widely used, X-rays are a suitable alternative to radioactive sources for luminescence dosimetry including radiofluorescence. The dual detection will allow us to isolate emissions from different mineral inclusions and directly correlate them to the elemental composition, thereby assessing the extent of overlap of the desired IR-RF emission centred at 880 nm and contaminating ones such as the possible unstable red emission associated with Fe³⁺.

Data availability

270 The SEM dataset and the original data used to produce µXRF maps are available online (Sontag-González et al., 2023).

Author contribution

MF, JT, RK and JLS designed the experiments and prepared the samples. RK and MF carried out the IR-RF measurements. SK organised and analysed the SEM EDS measurements. MF and JT carried out the μ XRF and μ XANES measurements. RK, MSG and MF analysed the results. MSG and RK prepared the manuscript with contributions from all authors. MF, JT and JLS obtained funding.

Competing interests

The authors declare that they have no conflict of interest.

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Figure 1: Illustration showing how contamination of the IR-RF signal can be removed by selecting only K-rich feldspar grains from sample X7343. The regenerative IR-RF curves were obtained from aliquots containing hundreds of unsorted grains (top) or 10–30 grains manually sorted into transparent shiny angular grains (middle) or white-pinkish rounded grains (bottom). The histogram shows the K-content determined by SEM-EDS for 100 grains (not measured for IR-RF). Representative examples of grains classified as K-rich and contaminating Fe-rich grains are shown.

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Figure 2: IR-RF dose-response curves of individual grains obtained after bleaching. Categories #1–#3 refer to grains with decreasing, increasing or no detectable signal, respectively (one category per row). The curves are shown (a, c, e) unnormalised and without background correction and (b, d, f) with intensities normalised to the signal maxima (defined as the median value of (b, f) the initial and (d) the final 20 channels) after subtracting as background the minimum signal of each grain (defined as the median value of (b, f) the final and (d) the initial 20 channels). For better visualisation, the normalised plots show the running median IR-RF with window of 7 values. The system background was determined as the mean value obtained from measuring five empty cups under the same conditions as the grains.







410 **Figure 3:** Simulated dose-response curves of theoretical aliquots varying the proportion of grains from categories #1 (desired decreasing signal) and #2 (increasing signal). The curves are the sum of two stretched exponentials using parameters obtained from fits of grains from samples Gi326 (category #1) and X3743 (category #2). The higher the signal contribution from category #2 grains, the more aberrant the sum curve becomes.







415 **Figure 4:** Bar charts of elements identified in µXRF spectra for grains in three categories, as exemplified in the insets: decreasing IR-RF signal (category #1), increasing IR-RF signal (category #2) or flat IR-RF signal indistinguishable from the background (category #3) during beta irradiation.







420 **Figure 5:** Ternary diagrams of relative XRF intensities attributed to (a) K, Fe and Pb, (b) Ba, Fe and Pb, and (c) K, Ca and Pb for grains of the three categories. Note that the contributions are not calibrated to mass or stoichiometry. The relative K contribution is, thus, not directly comparable to the K-feldspar K-content.







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Figure 6: Illustration showing three IR-RF curves obtained from three grains of sample X7368, classified as follows: decreasing IR-RF signal (category #1), increasing IR-RF signal (category #2) or flat IR-RF signal indistinguishable from the background (category #3) during beta irradiation. μXRF spectra were measured from the area bordered by red squares on the microscope images of the grains. The maps show the presence of K, Fe, and Pb on the same grains as the IR-RF curves. The elemental compositions are shown overlaid, with the colour scales normalised to the maximum contribution of each element for each grain.







430 **Figure 7:** μXANES spectra of Fe standards. The dashed vertical lines indicate the incident beam energies necessary to isolate emissions from specific oxidation states.



435 **Figure 8:** Maps of Fe oxidation states for a grain of sample X7368 (category #1; ID #17). Intensities are normalised to the maximum intensity of total Fe.





Table1: Overview of measured grains. Categories #1-#3 refer to grains with decreasing, increasing or no detectable signal, respectively.For two grains, two regions each were mapped, so we measured a total of 24 μ XRF maps.

Sample	Grain size (µm)	Number of measured grains		
		Category #1	Category #2	Category #3
Gi326	90–200	2	0	0
H22550	180–250	3	1*	2
X7343	180–255	0	1*	0
X7363	180–255	1	3	1
X7368	180–255	3	3	2
Total		9	8	5

*For these grains, two regions were mapped by μ XRF: the grain 'matrix' and an inclusion.